



CODEN [USA]: IAJ PBB

ISSN : 2349-7750

**INDO AMERICAN JOURNAL OF
PHARMACEUTICAL SCIENCES**

SJIF Impact Factor: 7.187

<https://doi.org/10.5281/zenodo.5564720>Available online at: <http://www.iajps.com>

Research Article

**FORMULATION AND EVALUATION OF MOUTH DISSOLVING
TABLETS OF METOCLOPROMIDE HCL USING NATURAL AND
SYNTHETIC SUPER DISINTEGRANTS****Abdul Sayeed¹, Dr.Pawan Kumar¹, Dr. Syed Areefulla Hussainy².**¹Research Scholar, School of pharmacy and medical science, Singhania University, Pacheri Bari,
Dist. Jhunjhunu-33315, Rajasthan (India)²Faculty of Pharmacy, Mesco College of Pharmacy, Hyderabad, Telangana State, India.**Article Received:** August 2021**Accepted:** September 2021**Published:** October 2021**Abstract:**

In the present study, formulation of mouth dissolving tablets for anti emetic drug using natural and synthetic super disintegrants by direct compression method. In the direct compression method, Plantago ovate mucilage (2-8%) was used as natural super disintegrants, Cross carmellose sodium (2-8%) was used as synthetic super disintegrants. Estimation of anti emetic drug in the prepared tablet formulations was carried out by extracting the drug with 0.1 N HCL and pH 6.8 phosphate buffer and measuring the absorbance at 272.6nm.

The prepared formulations were further tested for post compression parameters.

In vitro drug release pattern (0.1 N HCL and pH 6.8 phosphate buffer), stability study (at 40°C/ 75% RH for 6 months), drug-excipient interaction (IR spectroscopy) in vitro dispersion time respectively in comparison with formulations containing natural super disintegrant.

The promising formulations containing 8% w/w natural super disintegrant Plantago ovata mucilage emerged as the overall best formulation formulations (FM8).

Short-term stability studies of promising formulations indicated that there are no significant changes in drug content and in vitro dispersion time ($p < 0.05$). IR-spectroscopic studies indicated that there are no drug-excipient interactions. It can be concluded from the present work that natural super disintegrant were found to be superior in compression with synthetic super disintegrant with improved dissolution.

Keywords: Metoclopramide HCL, Plantago ovata mucilage, Cross carmellose sodium.

Corresponding author:**Abdul Sayeed,**

Research Scholar,

School of pharmacy and medical science,

Singhania University, pacheri bari,

Dist. Jhunjhunu-33315, Rajasthan (india).

E-mail: mohammedsayeed19@gmail.com

QR code



Please cite this article in press Abdul Sayeed et al, **Formulation And Evaluation Of Mouth Dissolving Tablets Of Metoclopramide Hcl Using Natural And Synthetic Super Disintegrants.**, Indo Am. J. P. Sci, 2021; 08(10).

1. INTRODUCTION:

Oral routes of drug administration have wide acceptance up to 50-60% of total dosage forms. Solid dosage forms are popular because of ease of administration, accurate dosage, self medication, pain avoidance and most importantly the patient compliance. One of the important drawbacks of this dosage forms for some patients, is the difficulty to swallow (Dysphasia) ¹⁻². Drinking water plays an important role in the swallowing of oral dosage forms. Often times people experience inconvenience in swallowing conventional dosage forms such as tablet when water is not available, in the case of the motion sickness and sudden episodes of coughing during the common cold, allergic condition and bronchitis. For this reason, tablets that can rapidly dissolve or disintegrate in the oral cavity have attracted a great deal of attention.

Mouth dissolving tablets are not only indicated for people who have swallowing difficulties, but also are ideal for active people.

MDTs are known by various names such as “fast-melting, fast-dissolving, oral disintegrating or orodispersible”. The European Pharmacopoeia defines the term “mouth dissolving” as a tablet that can be placed in the mouth where it disperses rapidly before swallowing. Fast dissolving tablets are those when put on tongue disintegrate instantaneously releasing the drug which dissolve or disperses in the saliva.

The faster the drug into solution, quicker the absorption and onset of clinical effect ³.

2. MATERIALS AND METHODS:

Metoclopramide HCL was procured from Karnataka antibiotics Pvt. Ltd., Bangalore; Sodium Starch Glycolate was procured from Vijlak Pharma Ltd., Mumbai, all other ingredients obtained from SD Fine Chemicals Pvt Ltd, Mumbai.

Preparation of calibration curve of Metoclopramide Hcl:

Procedure for Standard Curve in Distilled water: 100 mg of Metoclopramide HCl was dissolved in 100 ml of distilled water by shaking in volumetric flask (1000 µg/ml). 1 ml of this solution was taken and made up to 50 ml with distilled water, which gives 20 µg/ml concentration (stock solution).

From the stock solution, concentrations of 2, 4, 6, 8 and 10 µg/ml in distilled water were prepared. The absorbance of diluted solutions was measured at

272.6nm and a standard plot was drawn using the data obtained (figure no.1. spectrum of Metoclopramide HCL drug). The absorbance data of the concentrations are shown in table no.4.

Procedure for Standard Curve in pH 6.8 phosphate buffer:

100 mg of Metoclopramide HCL was dissolved in 100 ml of pH 6.8 phosphate buffer by shaking in volumetric flask (1000 µg/ml). 1 ml of this solution was taken and made up to 50 ml with pH 6.8 phosphate buffer, which gives 20 µg/ml concentration (stock solution). From the stock solution, concentrations of 2, 4, 6, 8 and 10 µg/ml in pH 6.8 phosphate buffer were prepared. The absorbance of diluted solutions was measured at 272.6nm and a standard plot was drawn using the data obtained (figure no.2. spectrum of Metoclopramide HCL drug). The absorbance data of the concentrations are shown in table no.5.

Isolation of Plantago ovate mucilage:

For the isolation of mucilage, seeds of *Plantago ovata* were used. They were soaked in distilled water for 48 hours and then boiled for 1 h for complete release of mucilage into water. The material collected was squeezed through muslin cloth for filtering and separating out the marc. Then, an equal volume of acetone was added to the filtrate so as to precipitate the mucilage. The separated mucilage was dried (in oven at temperature less than 60⁰), powdered, sieved (#60) and stored in a desiccator until further use ⁴.

FORMULATION DEVELOPMENT:

The best types of synthetic and natural superdisintegrants are incorporated in the formulation of MDTs like, Cross carmellose sodium and *Plantago ovate*. Before the tablet formulation the superdisintegrants was screened out and taken into formulation with other excipients for compression by direct compression method. The superdisintegrant shows good properties like, when the tablet comes in contact with liquid, it breaks up into smaller particles because of superdisintegrants are swells, hydrate, change the volume and produce a disruptive change in the tablet.

In this work, the direct compression method with aid of synthetic and natural superdisintegrants was attempted for the formulation development of mouth dissolving tablets of Metoclopramide hydrochloride. The Metoclopramide hydrochloride tablets are available in 5mg, 10mg doses in the market. Dose of 10mg is selected for the present study.

The development of the formulation of mouth dissolving tablets in the present study was mainly based on the type and concentration of synthetic and natural superdisintegrants. Synthetic and natural super disintegrates in different concentrations (2%, 4%, 6% and 8%) were used so as to get tablets with good physical properties. Ingredients like Microcrystalline cellulose and mannitol as directly compressible diluents, magnesium stearate and talc as lubricant, aerosil as flow promoter, aspartame as sweetening agent and pineapple flavor as enhance the palatability.

Preparation of powder blends of drug and excipients:

The powder blends for mouth dissolving tablets were prepared by taking ingredients given in Table no. 1. All the ingredients were passed through 60 mesh sieve separately and collected. Then ingredients were weighed and mixed in a geometrical order. First Microcrystalline cellulose, Mannitol and Super disintegrants were weighed and mixed together in glass mortar using a pestle. Then Drug and Aspartame were mixed and added in first mixer. Then Magnesium stearate, Talc and Aerosil were added and mixed. Finally flavor (Pineapple flavor) was added and mixed for 10-20 minutes.

Before tablets preparation, the mixture blends of all the formulations were subjected for compatibility studies (IR) and pre-compression parameter like Angle of repose, Bulk density, Tapped density, Percentage compressibility and Hausner ratio.

Preparation of Metoclopramide HCL Mouth dissolving tablets by direct compression:

Metoclopramide HCL mouth dissolving tablets were prepared in nine formulations FM0 to FM8 using the ingredients given in the Table no.1. Keeping the total weight of the tablet (150mg) kept constant in all the formulations. All the ingredients were passed through 60 mesh sieve separately and collected. Then ingredients were weighed and mixed in a geometrical order.

First microcrystalline cellulose, mannitol and super disintegrants were weighed and mixed together in glass mortar using a pestle. Then drug and aspartame were mixed and added in first mixer. The blend was then lubricated by mixing with magnesium stearate, talc and aerosil. Finally the mixture was blended with flavor. Then the powder blend was compressed. Tablets were prepared using 8 mm round flat-faced punches of the 16-station (Cadmach Machineries ltd.) rotary tablet compression machine. Compression force was kept constant for all formulation.

The mouth dissolving tablets were prepared and subjected to post compression parameters like hardness, friability, thickness, and weight variation, *In-vitro* dispersion time, wetting time, water absorption ratio, drug content, *In-vitro* disintegration time and *In-vitro* dissolution.

Table no. 1: Formulation of Metoclopramide HCL Mouth Dissolving Tablets

S.No	Ingredients (mg/tab)	FM0	FM1	FM2	FM3	FM4	FM5	FM6	FM7	FM8
1	Metoclopramide HCL	10	10	10	10	10	10	10	10	10
2	Cross carmellose sodium	---	3	6	9	12	---	---	---	---
3	Plantago ovata mucilage	---	---	---	---	---	3	6	9	12
4	Microcrystalline cellulose	50	50	50	50	50	50	50	50	50
5	Aspartame	5	5	5	5	5	5	5	5	5
6	Magnesium stearate	1.5	1.5	1.5	1.5	1.5	1.5	1.5	1.5	1.5
7	Talc	1.5	1.5	1.5	1.5	1.5	1	1	1	1
8	Aerosil	1.5	1.5	1.5	1.5	1.5	1.5	1.5	1.5	1.5
9	Pineapple flavour	0.5	0.5	0.5	0.5	0.5	1	1	1	1
10	Mannitol	80	77	74	71	68	78	75	72	69
	TOTAL	150	150	150	150	150	150	150	150	150

PRE-COMPRESSION ASSESSMENT OF POWDER BLEND⁵

Different parameters were evaluated for prepared powder blend using following methods.

Angle of repose

Angle of repose is defined as maximum angle possible between the surface of the pile of powder and the horizontal plane.

The friction force in a loose powder can be measured by the angle of repose (θ). It is an indicative of the flow properties of the powder.

The angle of repose is calculated by using fixed funnel method. In this method the funnel was fixed to a stand at definite height (h). The graph paper was placed on a flat horizontal surface. Then powder blend was allowed to fall freely on the paper through the funnel, until the apex of the conical pile just touches the tip of the funnel. The height and radius of pile was noted and from this angle of repose was determined with the help of given formula.

The formula for calculating angle of repose is

$$\theta = \tan^{-1} (h / r)$$

$$\tan (\theta) = h / r$$

Bulk Density

Bulk density is the ratio of total mass of powder to the bulk volume of powder. It was measured by pouring the accurately weighed 2g of powder blend (passed through 20 mesh sieve) was placed in a 10ml graduated measuring cylinder. And then initial volume was observed, this initial volume is called as bulk volume.

From this the bulk density was calculated by using the following formula.

Bulk-density = Mass of the powder/Bulk volume.

Tapped Density

Tapped density is the ratio of total mass of powder to the tapped volume of powder. Accurately weighed amount of powder blend was placed in a measuring cylinder and the volume was measured by tapping of powder for 500 times and the tapped volume was noted. The tapped density was calculated by using following formula.

Tapped-density= Mass of the powder/Tapped volume.

Compressibility Index

Compressibility index is indicates the powder flow properties.

It is expressed in percentage. Compressibility index is based on the bulk density and tapped density, the percentage compressibility of the powder blend was determined by using the following formula.

$$\text{Carr's Index} = \frac{\text{Tapped-density}-\text{Bulk density}}{\text{Tapped density}} \times 100$$

Hausner's Ratio

Hausner ratio is an indirect index of ease of powder flow. It was calculated by the following formula.

$$\text{Hausner's ratio} = \frac{\text{Tapped density}}{\text{Bulk density}}$$

POST-COMPRESSION ASSESSMENT OF POWDER BLEND⁶⁻⁹

Thickness:

The thickness of the tablets was determined by using Digital vernier Calipers. Thickness mainly depends upon the die filling, physical properties of material to be compressed under compression force. Three tablets were randomly taken from each formulation, mean and standard deviation values were calculated. It is expressed in mm.

Hardness

The Monsanto hardness tester was used to determine the tablet hardness. The tablet was held between a fixed and moving jaw. Scale was adjusted to zero load was gradually increased until the tablet fractured. The value of the load at that point gives a measure of hardness of the tablet. Three tablets were randomly taken from each formulation, mean and standard deviation values were calculated. It is expressed in kg/cm^2 .

Friability

The friability test for tablets was performed to assess the effect of abrasion and shocks. Roche friabilator was used for the percent friability of the tablets. This device subjects the tablet to the combined effect of abrasion and shock in a plastic chamber revolving at 25 rpm and dropping a tablet at a height of 6 inches in each revolution. Pre-weighted sample of tablets was placed in the friabilator and were subjected to the 100 revolutions. Then the tablets were removed and de dusted by using a soft muslin cloth and reweighed. The weight lost should not exceed the limit 1.0%. The percentage friability was measured by using the following formula.

$$\% \text{Friability} = \frac{\text{Initial Weight} - \text{Final weight}}{\text{Initial Weight}} \times 100$$

Weight Variation

The weight variation test was performed as per I.P. Twenty tablets were randomly selected from each batch and individually weighed. And then average weight was calculated from the total weight of all tablets. The individual weights were compared with the average weight. The tablets pass the test for weight variation test if no more than two tablets are outside the percentage limit and if no tablet differs by more than two times the percentage limit.

***In-vitro* Dispersion Time**

In-vitro dispersion time was measured by dropping a tablet into a Petridis containing 10ml of phosphate buffer pH 6.8 solutions at $37 \pm 0.5^\circ\text{C}$. Three tablets from each batch were randomly selected and tested the time required for complete dispersion of a tablet was measured. The *in-vitro* dispersion time is expressed in seconds.

Wetting Time

A piece of tissue paper folded double was placed in a Petri dish (6.5cm) containing 6 ml of water. the tablet was placed on the paper, and the time for complete wetting of the tablet was measured in seconds. The method was slightly modified by maintaining water at 37°C . Wetting time corresponding to the time taken for the tablet to disintegrate when kept motionless on the Petri dish.

Water Absorption Ratio

A piece of tissue paper folded twice was placed in a Petri dish (6.5cm) containing 6 ml of water. A tablet was put on the tissue paper and the time required for the complete wetting was measured. The wetted tablet was then weighed. Water absorption ratio, R, was determined using following equation.

$$R = 100 (W_a - W_b) / W_b$$

Where,

W_a = Weight of the tablet after absorption.

W_b = Weight of the tablet before absorption.

Disintegration Time

The process of breakdown of a tablet in to a smaller particle is called as disintegration. The *in-vitro* disintegration time of a tablet was determined using disintegration apparatus as per I.P specifications.

I.P specifications: Place one tablet in each of 6 tubes of the basket. Add a disc to each tube and run the apparatus using pH 1.2 maintained at $37 \pm 2^\circ\text{C}$ as the immersion liquid. The assembly should be raised and lowered between 30 cycles per minute in the pH 1.2

maintained at $37 \pm 2^\circ\text{C}$. The time in seconds taken for complete disintegration of the tablet with no palpable mass remaining in the apparatus was measured and recorded.

Drug content

Three tablets weighed and crushed in a mortar then weighed powder contain equivalent to 10mg of drug transferred in 100ml of phosphate buffer pH 1.2. Subsequently, the solution in volumetric flask was filtered, suitable dilutions will be carried out. And final solution was analyzed at 272.6nm using UV-visible spectrophotometer Shimadzu UV- 2450, Japan.

***In-vitro* Dissolution Studies of Metoclopramide HCL:**

In vitro dissolution of mouth dissolving tablets were studied in USP type-II dissolution apparatus (Electrolab) employing a paddle stirrer. 900 ml of phosphate buffer P^{H} 6.8 was used as dissolution medium. The stirrer was adjusted to rotate at 50 rpm. The temperature of dissolution medium was maintained at $37 \pm 0.5^\circ\text{C}$ throughout the experiment. One tablet was used in each test. Samples of dissolution medium (5 ml) were withdrawn by means of syringe fitted with pre-filter at known intervals of time and analyzed for drug release by measuring the absorbance at 272.6nm. The volume withdrawn at each time interval was replaced with fresh quantity of dissolution medium. Cumulative percent released was calculated and plotted against time. The results are given in table no.9 to 11 and figure no.3 to 5.

KINETIC STUDY¹⁰

The results of *in vitro* release profile obtained for all the formulations were plotted in modes of data treatment as follows:

Zero order kinetic model:

Cumulative % drug released versus time.

First order kinetic model:

Log cumulative percent drug remaining versus time.

Higuchi model:

Cumulative percent drug released versus square root of time.

Korsmeyer equation / Peppas model:

Log cumulative percent drug released versus log time.

Zero order kinetics:

Zero order release would be predicted by the following equation:

$$A_t = A_0 - k_0 t$$

Where,

A_t = Drug release at time 't'

A_0 = Initial drug concentration.

k_0 = Zero-order rate constant (hr^{-1})

When the data is plotted as cumulative percent drug release versus time, if the plot is linear then the data obeys Zero – order kinetics and its slope is equal to Zero order release constant k_0 .

First Order Kinetics:

First – order release could be predicted by the following equation:

$$\log C = \log C_0 - Kt / 2.303$$

Where,

C = Amount of drug remained at time 't'

C_0 = Initial amount of drug.

K = First - order rate constant (hr^{-1})

When the data plotted as log cumulative percent drug remaining versus time, yields a straight line, indicating that the release follows first order kinetics. The constant 'K1' can be obtained by multiplying 2.303 with the slope value.

Higuchi's Model:

Drug release from the matrix devices by diffusion has been described by following Higuchi's classical diffusion equation:

$$Q = [DC / t (2A - Cs) Cst]^{1/2}$$

Where,

Q = Amount of drug release at time 't'

D = Diffusion coefficient of the drug in the matrix. A

= Total amount of drug in unit volume of matrix.

C_s = Solubility of drug in the matrix.

C = Porosity of the matrix.

t = Tortuosity.

T = Time (hrs at which q amount of drug is released).

Above equation can be simplified as if we assume

that 'D', 'Cs' and 'A' are constant. Then equation becomes.

When the data is plotted according to equation i.e. cumulative drug release versus square root of time yields a straight line, indicating that the drug was released by diffusion mechanism. The slope is equal to 'K'.

$$Q = kt^{1/2}$$

Korsmeyer Equation / Peppas Model:

To study the mechanism of drug release from the liposomal solution, the release data was also fitted to the well-known exponential equation which is often used to describe the drug release behavior from polymeric systems.

$$M_t / M_\infty = K t^n$$

Where,

M_t / M_∞ = The fraction of drug released at time 't'.

K = Constant incorporating the structural and geometrical characteristics of the drug polymer system.

n = Diffusion exponent related to the mechanism of the release.

Above equation can be simplified as follows by applying log on both sides,

$$\log M_t / M_\infty = \log K + n \log t$$

STABILITY STUDIES¹¹

Stability is defined as "the capacity of the drug product to remain within specifications established to ensure its identity, strength, quality and purity" (FDA 1987).

Stability studies of pharmaceutical products were done as per ICH guidelines. These studies are designed to increase the rate of chemical or physical degradation of the drug substance or product by using exaggerated storage conditions. Basically, there are two types of stability studies:

Short -term stability studies**Long- term stability studies**

Table no: 2: Stability conditions according to ICH guidelines

Types	Conditions		Minimum time period at submission (month)
	Temperature ($^{\circ}\text{C}$)	Relative humidity(%)	
Short-term testing	40 ± 2	75 ± 5	6
Long-term testing	25 ± 2	60 ± 5	12

Method:

Selected formulations were stored at different storage conditions at elevated temperatures such as $25^{\circ}\text{C} \pm 2^{\circ}\text{C}$ / $60\% \pm 5\%$ RH, and $40^{\circ}\text{C} \pm 2^{\circ}\text{C}$ / $75\% \pm 5\%$ RH for 90 days. The samples were withdrawn at intervals of 30 days and checked for physical changes, hardness, friability, drug content and percentage drug release.

PHYSICOCHEMICAL EVALUATION OF DRIED POWDERED MUCILAGE¹²⁻¹⁴

The following physicochemical tests for mucilage

Organoleptic properties: Organoleptic properties such as physical appearance, colour, odour and taste of dried powdered mucilage were determined.

Solubility test: The solubility of dried powdered mucilage was determined by adding a pinch in the solvent such as water.

Total ash: Total ash was determined on 1 gm of

dried powdered mucilage.

Loss on drying: Loss on drying was determined for an appropriate quantity of dried powdered mucilage at 105°C for 5 hours.

$$\text{LOD (\%)} = (\text{Wt of water in sample} / \text{Wt of dry sample}) \times 100$$

Swelling factor: Swelling factor was determined by putting 1 gm of the drug in the measuring cylinder (25 ml capacity) in 20 ml water with occasional shaking. The volume occupied by the seeds after 24 hours of wetting is measured.

Flow properties of dried mucilage powder: The flow properties of dried mucilage powder such as Angle of repose, Bulk density, Tapped density, Carr's index and Hausner's ratio were determined.

TABLE NO. 3: PHYSICOCHEMICAL TESTS FOR MUCILAGE

S.No	Physico chemical parameters	Plantago ovata mucilage
1	Solubility	Slightly soluble in water
2	Loss on drying (%)	10 ± 0.011
3	Swelling ratio	9 ± 0.145
4	Total ash (%)	4 ± 0.021
5	Angle of repose	$26.56^{\circ} \pm 0.251$
6	Bulk density g/cm^3	0.42 ± 0.055
7	Tapped density g/cm^3	0.46 ± 0.085
8	Carr's index (%)	10.03 ± 0.012
9	Hausners ratio	1.08 ± 0.056

All parameters (\pm SD) n=3

FTIR Spectroscopy:

The interaction between drug and excipients was studied by using FTIR spectroscopy. In the preparation of tablet formulation, drug and excipients may interact as they are in close contact with each other, which could lead to the instability of drug. Preformulation studies regarding the drug-excipients interaction are therefore very critical in selecting appropriate polymer. FTIR spectroscopy was employed to ascertain the compatibility between drug and the selected excipients. Potassium bromide, pure drug and the excipients were heated to 105°C for one hour in a hot air oven to remove the moisture content. Then in presence of IR lamp, potassium bromide was mixed with drug and or excipients and the spectra were taken. FTIR spectrum of drug was compared with FTIR spectra of excipients.

3. RESULT AND DISCUSSION:**Determination of λ_{max} and standard Calibration Curve of Metoclopramide HCL in distilled water:**

100 mg of Metoclopramide HCL was dissolved in 100 ml of distilled water by shaking in volumetric flask (1000 $\mu\text{g/ml}$). 1 ml of this solution was taken and made up to 50 ml with distilled water, which gives 20 $\mu\text{g/ml}$ concentration (stock solution).

From the stock solution, concentrations of 2, 4, 6, 8 and 10 $\mu\text{g/ml}$ in distilled water were prepared. The absorbance of diluted solutions was measured at 272.6nm and a standard plot was drawn using the data obtained. The correlation coefficient was calculated. The absorbance data of the concentrations are shown in table-4.

Table-4: Standard graph of Metoclopramide HCL in Distilled water (λ_{max} 272.6)

Concentrations $\mu\text{g/ml}$	Absorbance			
	I	II	III	Mean \pm SD
0	0.000	0.000	0.000	0.000 \pm 0.000
2	0.085	0.087	0.086	0.086 \pm 0.001
4	0.174	0.161	0.164	0.166 \pm 0.006
6	0.248	0.241	0.254	0.248 \pm 0.006
8	0.334	0.328	0.330	0.330 \pm 0.003
10	0.409	0.424	0.412	0.415 \pm 0.007

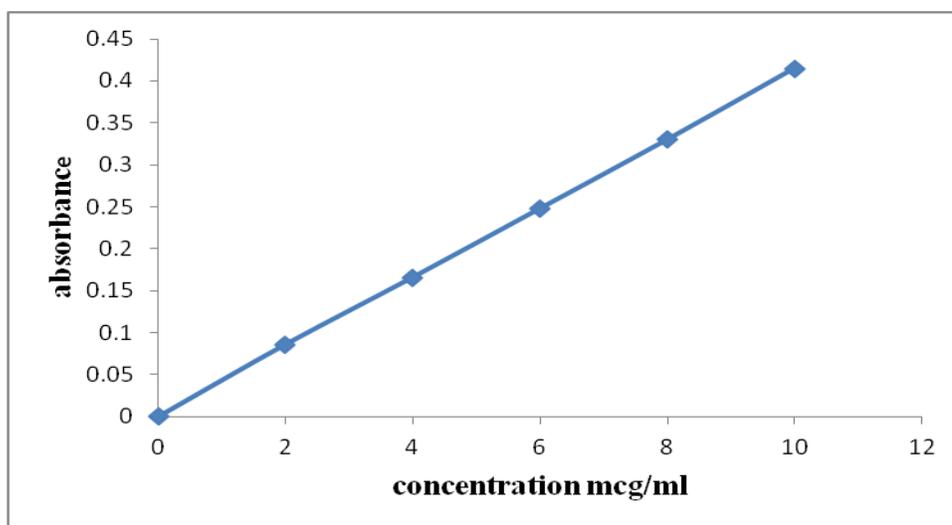


Figure-1: Standard graph of Metoclopramide HCL in Distilled water (λ_{max} 272.6)

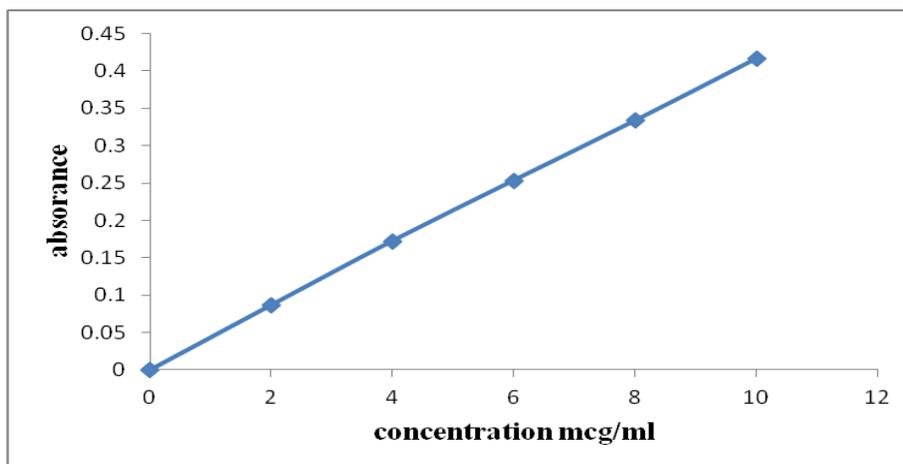
Determination of λ max and standard Calibration Curve of Metoclopramide HCL in pH 6.8 phosphate buffer:

100 mg of Metoclopramide HCL was dissolved in 100 ml of pH 6.8 phosphate buffer by shaking in volumetric flask (1000 $\mu\text{g/ml}$). 1 ml of this solution was taken and made up to 50 ml with pH 6.8 phosphate buffer, which gives 20 $\mu\text{g/ml}$ concentration (stock solution).

From the stock solution, concentrations of 2, 4, 6, 8 and 10 $\mu\text{g/ml}$ in pH 6.8 phosphate buffer were prepared. The absorbance of diluted solutions was measured at 272.6nm and a standard plot was drawn using the data obtained. The correlation coefficient was calculated. The absorbance data of the concentrations are shown in table-5.

Table-5: Standard graph of Metoclopramide HCL in pH 6.8 phosphate buffer(λ_{max} 272.6nm)

Concentration' $\mu\text{g/ml}$	Absorbance			
	I	II	III	Mean \pm SD
0	0.000	0.000	0.000	0.000 \pm 0.000
2	0.084	0.087	0.090	0.087 \pm 0.003
4	0.170	0.174	0.176	0.173 \pm 0.003
6	0.245	0.257	0.260	0.254 \pm 0.007
8	0.342	0.348	0.346	0.334 \pm 0.003
10	0.422	0.428	0.431	0.417 \pm 0.004

**Figure-2: Standard graph of Metoclopramide HCL in pH 6.8 phosphate buffer(λ_{max} 272.6 nm)****Pre-compression parameters of Metoclopramide powder blend:**

Powder blend for direct compression containing drug and various excipients were subjected for pre compression parameters (micromeritic properties) to study the flow properties of powder blend to achieve uniformity of tablet weight.

The bulk density of powder blend was found to be in the range of 0.27 to 0.37 g/cc, tapped density was found to be in the range of 0.36 to 0.54 g/cc, angle of

repose was found to be in the range of 27.16 to 32.11°, Carr's index was found to be in the range of 10.75% to 19.88%, Hausner's ratio was found to be in the range of 1.11 to 1.19. All the formulations show good results and lies within the acceptable range which indicate good flow properties.

The results of all the pre compression parameters are given in table no.6.

Table no 6: Pre-compression parameters of Metoclopramide HCL powder blend FM0 - FM8

Formulation code	Bulk Density (g/cc)	Tapped density (g/cc)	Angle of repose (degree)	Carr's index (%)	Hausner's ratio
FM0	0.33	0.36	30.19	10.75	1.11
FM1	0.34	0.45	27.16	19.88	1.18
FM2	0.29	0.39	28.19	19.76	1.19
FM3	0.27	0.46	27.98	16.18	1.18
FM4	0.29	0.43	28.88	14.28	1.18
FM5	0.29	0.54	31.14	15.38	1.19
FM6	0.37	0.38	32.11	14.93	1.16
FM7	0.29	0.46	30.19	16.29	1.18
FM8	0.27	0.44	29.27	17.44	1.17

All results expressed as mean \pm SD, n = 3

Post Compression parameters of Metoclopramide HCL Mouth dissolving tablets:

All the tablet formulations were subjected for organoleptic, physical and chemical evaluation. Shape and colour, Weight Variation, Thickness, Hardness, Friability, Drug Content, Wetting time, Water absorption ratio, Disintegration time, *In vitro* dispersion time, and *In-vitro* drug release studies were carried out.

Appearance of the tablets: Tablets were selected randomly from each formulation batch and examined under lens for shape and in presence of light for colour. Tablets showed concave, circular shape in white color and all tablets showed very good appearance without any capping or lamination and found satisfactory.

Thickness: Thickness of all the formulations were found in the range between 2.11 ± 0.04 mm to 2.22 ± 0.01 mm and summarized in table no.7.

Weight Variation: The percent Weight Variation of all the formulations were summarized in table. All the tablets were passed weight variation test as the % variation was within the pharmacopoeial limits of 7.5%. It was found to be from 146.4 ± 1.69 to 150.5 ± 0.58 , the weight of the all tablets was found to

be uniform due to good flow property and compressibility of all the formulations.

Hardness: The hardness of tablets was tested using Pfizer hardness tester to find out whether they could retain their physical shape or not. The hardness of all the tablets was found to be in the range of 2.81 ± 0.1 kg/cm² to 2.99 ± 0.5 kg/cm² and the results were summarized in table no.7.

Friability: Tablet strength was tested by Roche Friabilator and the tablets of all formulations showed very good friability with less than **0.53%** which is well and within wide accepted range of Pharmacopoeia limit (1.0%) and results were given in table no.7.

Drug Content uniformity: The drug content uniformity was performed for all the formulations, the mean value and standard deviation of all the formulations were calculated, the low values of standard deviation indicates uniform drug content within the tablets. The percent drug content of all the tablets was found to be in the range of 99.28 ± 1.52 to 101 ± 2.02 percent (which was within the acceptable limits of $\pm 5\%$) and results were given in table no. 7.

Table no 7: Post Compression parameters of formulations FM0 – FM8

Formulation code	Weight Variation *	Thickness*	Hardness**	Friability**	Drug Content**
FM0	149.8±1.61	2.11±0.04	2.86±0.5	0.49	99.28±1.52
FM1	148.5±2.54	2.13±0.01	2.18±0.5	0.55	101±1.09
FM2	147.3±2.21	2.22±0.01	2.96±0.5	0.54	99.45±2.11
FM3	150.0±1.49	2.14±0.12	2.91±0.1	0.49	99.45±1.01
FM4	148.4±1.89	2.16±0.06	2.91±0.1	0.51	101.0±1.57
FM5	150.5±0.58	2.22±0.01	2.86±0.5	0.54	99.28±1.52
FM6	147.1±1.14	2.14±0.12	2.81±0.1	0.49	99.70±1.14
FM7	149.5±2.12	2.16±0.06	2.83±1.4	0.51	100±1.57
FM8	146.4±1.69	2.12±0.02	2.99±0.5	0.42	101±2.02

All results expressed as mean ± SD, n = 3

Water absorption ratio:

The water absorption ratio of all the formulations was found to be 45±1% to 85.11±1.11 %. The results were depicted in Table.no.8.

Disintegration-Time:

The disintegration time of all the formulations was found to be 17±1.12 sec to 299±1.62 sec. The results were depicted in Table.no.8.

Wetting time:

The Wetting time of all the formulations was found to be 13.18±1.5 sec to 294±1.62 sec. The results were depicted in Table.no.8.

In vitro dispersion time:

The *In vitro* dispersion time of all the formulations was found to be 21.11±0.15 sec to 99±2 sec. The results were depicted in Table.no.8.

Table no 8: Post compression parameters of formulations FM0 – FM8

Formulation Code	Wetting time (sec)	Water absorption ratio (%)	Disintegration time (sec)	<i>In vitro</i> dispersion time (sec)
FM0	294±1.62	45±1	299±1.62	99±2
FM1	43.11±1.0	60.22±3.8	88±0.34	56.03±2.47
FM2	32.26±0.7	63.21±1.5	71±0.11	47.0± 2.10
FM3	29.33±1.52	70.75±1.01	47±0.29	38.42± 1.90
FM4	25.99±1.5	82.12±1.14	34±0.12	29.34± 0.70
FM5	39.64±2.08	68.12±1.61	69±1.55	42.66±1.52
FM6	26.32±1.01	76.46±2.9	56±1.82	37.66±1.52
FM7	20.19±1.12	80.46±2.9	37±2.05	32.33±2.51
FM8	13.18±1.5	85.11±1.11	17±1.12	21.11±0.15

All results expressed as mean ± SD, n = 3

***In-vitro* drug release studies**

Tablets containing Metoclopramide HCL were studied for *In-vitro* drug release studies as per the procedure described in methodology. All formulations were subjected for dissolution studies. The samples were withdrawn at specified time intervals and analyzed by UV-Visible Spectrophotometer at 272.6 nm.

Drug release profile was studied using percentage drug release versus time (hr) plot. The results were depicted in Table No.9 to 11 and figure no 3 to 5. Formulations FM0, FM1, FM2, FM3 and FM4 showed 37.84±0.6 %, 63.08±2.78%, 69.69±2.78, 80.37±1.52% and 92.69±1.54%. Release of drug respectively at 30min. Formulations FM5, FM6, FM7 and FM8 showed 75.08±2.78%, 80.69±2.78, 90.37±1.52% 98.25±1.65% respectively at 30 minutes.

Among all formulations FM8 containing 8% plantago ovate mucilage as natural super disintegrant was found to be promising and has shown faster release of drug.

Table no 9: *In -Vitro* drug release characteristics of Metoclopramide HCL without Superdisintegrants (FMO)

Time (min)	Cumulative % of drug release without Superdisintegrant (FMO)
0	0
05	10.97±1.4
10	15.27±0.5
15	26.67±1.2
20	29.29±1.4
25	32.71±1.9
30	37.84±0.6

All results expressed as mean ± SD, n = 3

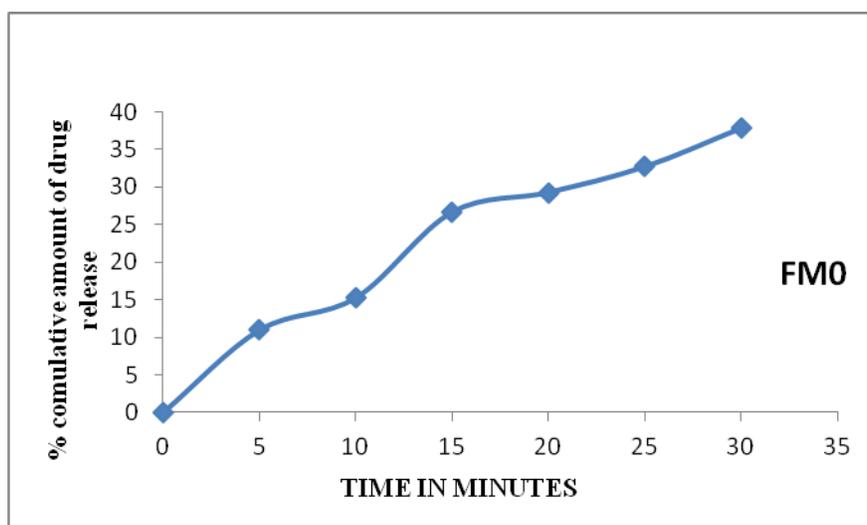


Figure. No.3. %Cumulative amount of drug release Vs Time of FM0.

Table no 10: *In –Vitro* drug release characteristics of Metoclopramide HCL with Croscarmellose sodium (FM1– FM4)

Time (Min)	Cumulative % of drug release with Croscarmellose sodium			
	FM1	FM2	FM3	FM4
0	0	0	0	0
05	23.34±1.00	26.75±1.54	34.05±0.54	39.55±1.24
10	28.70±1.34	38.20±1.43	45.92±1.37	56.83±2.04
15	39.39±2.01	47.16±2.17	58.23±2.05	66.80±1.51
20	47.50±2.67	58.92±2.53	67.73±0.84	73.77±1.58
25	59.22±1.45	65.70±1.73	76.80±1.54	78.52±1.05
30	63.08±2.78	69.69±2.78	80.37±1.52	92.69±1.54

All results expressed as mean ± SD, n = 3

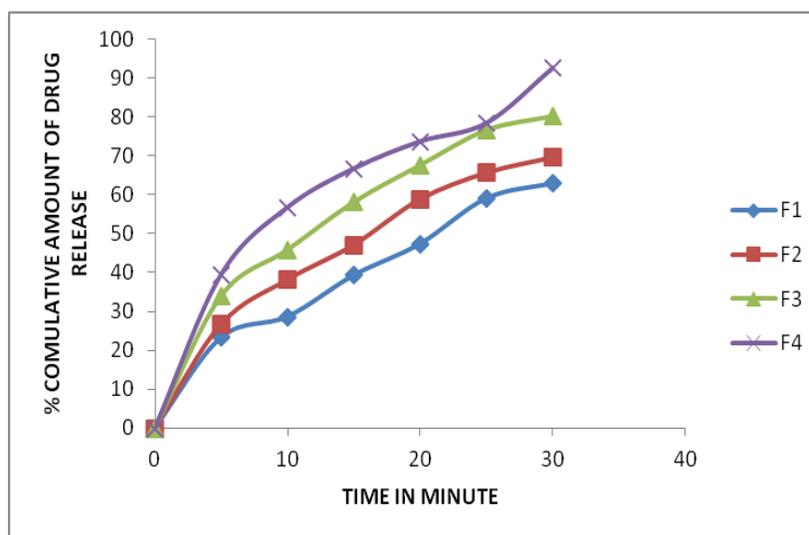


Figure. No.4. %Cumulative amount of drug release Vs Time of FM1 to FM4.

Table no 11: *In-Vitro* drug release characteristics of Metoclopramide HCL with Plantago ovate mucilage. (FM5– FM8)

Time (Min)	Cumulative % of drug release with Plantago ovate mucilage.			
	FM5	FM6	FM7	FM8
0	0	0	0	0
05	29.08±1.73	34.37±1.56	37.42±1.02	39.82±1.65
10	35.54±2.74	44.37±2.34	48.01±1.39	54.78±2.64
15	46.26±2.46	53.91±2.68	59.32±1.75	66.83±2.73
20	57.98±2.39	69.47±2.47	76.89±1.91	79.05±2.78
25	66.32±1.87	75.65±1.47	80.92±2.36	86.17±2.18
30	75.08±2.78	80.69±2.78	90.37±1.52	98.25±1.65

All results expressed as mean ± SD, n = 3

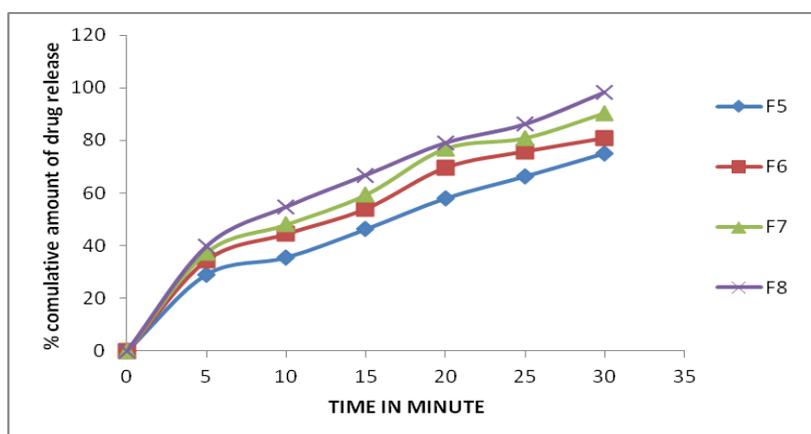


Figure. No.5: %Cumulative amount of drug release Vs Time of FM5 to FM8.

STABILITY STUDIES:

Short-term stability studies conducted on formulation (FM8) at 40°C/ 75% RH for 3 months have shown no significant changes in physical appearance, drug content and *in vitro* dispersion time and dissolution and results were summarized in table 12.

Table no 12: Stability data for formulation Metoclopramide HCL (FM8)

Name of Test	Initial	1 st month	2 nd month	3 rd month	6 th month
Physical Changes	No changes	No changes	No changes	No changes	No changes
Dissolution					
05 minutes	39.55±1.24	39.85±1.22	39.89±1.20	39.95±1.21	38.99±1.19
10 minutes	53.83±2.04	53.83±2.14	53.83±2.14	53.83±2.14	53.80±2.14
15 minutes	68.80±1.51	68.80±1.50	68.80±1.50	68.80±1.50	68.80±1.50
20 minutes	79.77±1.58	79.77±1.57	79.77±1.57	79.77±1.57	79.77±1.57
25 minutes	88.52±1.05	88.52±1.05	88.52±1.05	88.52±1.05	88.52±1.05
30 minutes	98.25±1.65	98.15±1.65	98.15±1.65	98.15±1.65	98.15±1.65
Assay (%)	98.25±1.65	98.15±1.11	98.11±1.10	98.11±1.10	98.11±1.10
Friability (%)	0.44	0.44	0.42	0.42	0.42
Disintegration (Sec)	17±1.12	17±1.12	17±0.99	17±0.75	17±0.15
Dispersion time (Sec)	21.11±0.15	21.11±0.19	21.11±0.21	21.11±0.26	20.11±0.29

All results expressed as mean ± SD, n = 3

KINETICS STUDIES:

The *in-vitro* drug release data of the fast-dissolving tablets were evaluated kinetically, by Zero order, First order, Higuchi, Peppas. The data were processed for regression analysis using PCP DISSO V3 Software. The regression coefficient (R) value for Zero order, first order, Higuchi, Peppas, for all the formulations were shown in Table. No 13. The formulations FM8 follows zero order kinetics. The release of drug may be depending on disintegration time.

TABLE NO 13: KINETIC STUDIES OF METOCLOPROMIDE HCL MOUTH DISSOLVING TABLET:

Formulation code	Zero order (R ² value)	First order (R ² value)	Higuchi (R ² value)	Korsmeyer-Peppas (R ² value)
FM0	0.9554	0.9671	0.9494	0.9589
FM1	0.9868	0.9790	0.9895	0.9802
FM2	0.9777	0.9929	0.9862	0.9678
FM3	0.9761	0.9942	0.9889	0.9842
FM4	0.9647	0.8887	0.9595	0.9280
FM5	0.9950	0.9775	0.9984	0.9967
FM6	0.9768	0.9845	0.9770	0.9460
FM7	0.9791	0.9598	0.9778	0.9458
FM8	0.9909	0.8216	0.9942	0.9904

4. CONCLUSION:

- Mouth dissolving tablets of Metoclopramide HCL were successfully formulated by employing direct compression method, using natural and synthetic Super disintegrants.
- Firstly extraction of plantago ovata mucilage used as a natural super disintegrating agents. .
- The physicochemical parameters like pre-compression and post-compression evaluation were performed as per pharmacopeia standards and compatibility study was done by FTIR method.
Based on the above studies, following conclusions can be drawn.
- The FTIR studies indicated that the drug was compatible with the carriers, polymers and other excipients used in the dosage form.
- Pre-compression parameter results showed good flow properties.
- Mouth dissolving tablets of Metoclopramide HCL were prepared by direct compression method.
- Croscarmellose Sodium used as synthetic super disintegrants.
- Magnesium stearate is used as a lubricant. Talc is used as a glidant.
- Aspartame is used as sweetening agent.
- Post-Compression parameter results found to be optimum. Thus hardness of the tablets shown sufficient to withstand the shock. All the formulations tablets were found uniformity in weight.

- The drug content was uniform in all the tablet formulations indicating uniform distribution of drug within the matrices.
- Based on the *in-vitro* disintegration time and dissolution studies of Metoclopramide HCL formulations FM8 containing Superdisintegrant as seeds of plantago ovata mucilage were found to be promising and showed a disintegration time 17±1.12 sec and drug release profile 98.25±1.65 respectively, when compared to the synthetic and other natural super disintegrant.
- The formulations subjected for kinetic studies and shown zero order kinetics.
- The stability studies carried out as per ICH guidelines for 3 months. Results showed that the formulations were stable and intact without any interaction.
- Finally, it was concluded that the MDTs of Metoclopramide HCL formulations containing Superdisintegrant as seeds of plantago ovata mucilage showed less disintegration time and *in-vitro* drug release study faster than the synthetic super disintegrant.
- Formulations were found to be complying with all the properties of tablets and the formulations were satisfactory.

5. REFERENCES:

1. Shah S, Madan S and Agrawal SS. Formulation and evaluation of microsphere based oro dispersible tablets of Itopride HCl. DARU J pharmsci, 20(24) (2012) 1-12.

2. Bharathi A, Ramakrishna V. Formulation development and in-vitro evaluation of orally disintegrating tablets of Amlodipine besylate. *IJRPC*, (2012) 1029-34.
3. Debjit Bhowmik, Chiranjib.B, Krishnakanth, Pankaj, Margret Chandra.R. Fast dissolving tablets. *Journal of Chemical & Pharmaceuticeutical Research*, 2009; 1(1): 163-177.
4. Malla VC, Mani Kumari M, Mariyadas U, Sushma M, Mounika N. Formulation and Evaluation of Ibuprofen Tablets by using different superdisintegrating agents. *World J Pharm Pharm Sci* 2016;5(5):875-883.
5. Dali shukla, Subhashis Chakraborty, Sanjay Singh, Brahmeshwarn Mishra.Mouth dissolving tablets: an overview of formulation technology.2009, 77:309- 326.
6. Tapan kumar, Dulal Krishna, Rana.Formulation aspects in the development of Orodispersible tablets: an overview. *International journal of pharmacy and Pharmaceutical sciences*, 2010: May 07:2(3).
7. Tejvir, Bhavandeep , Sandeep ,Gupta GD. Mouth dissolving tablets; A novel approach to drug delivery. *International journal of current pharmaceutical research*.2011:July 17:3(1).
8. Pagar R, Ahirrao S, Yallatkar T, Wagh M. Rev on orodispersible tablets. *Int J Res Dev Pharm L Sci* 2014;3:949-58.
9. Mishra S, Prajapati K, Bhardwaj P. A rev on formulation and evaluation for mouth dissolving tablet. *World J Pharm PharmSci* 2014;8:1778-810.
10. Kuchekar S, Badha C, Mahajan S. Mouth dissolving tablets. A novel drug del system. *Pharm times* 2003; 35:7-9.
11. Kumaran K, Sreekanth J, Palanisamy S. Formulation, development and evaluation of Levodopa-Carbidopa orally disintegration tablets. *J Chem Pharm Res* 2011; 3(3):169-175.
12. Belet H, Derle V. Analysis of patents pertaining to superdisintegrants used in tablet manufacturing. *J intellectual Property Rights* 2008;13:601-604.
13. Omidian H, Park K. Swelling agents and devices in oral drug delivery. *J Drug Del Sci Tech* 2008;18(2):83-93.
14. Bhardwaj S, Jain V, Sharma S, Jat RC, Jain S. Orally disintegrating tablets. A rev *Drug Invention Today* 2010;2(1):81-88.