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Research Article

**FORMULATION AND EVALUATION OF NATURAL ANTI-
ACNE SERUM USING CINNAMOMUM CAMPHORA
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Abstract:

The moisture content present in human skin makes it look young and the use of moisturizer results in fastening the moisture with a surface film of oil. Acne vulgaris is one of the most commonly seen diseases among the youth. The present study is focused on the use of herbs as moisturizer for acne treatment. The anti-acne moisturizer was formulated from herbal crude extracts and investigated the physico-chemical parameters as well as antibacterial activity of the formulation. The study revealed that extract of Cinnamomum Camphora possessed the potential for inhibiting acne. It was observed that the optimal formula of anti-acne moisturizer was satisfactorily effective to control acne inducing bacteria i.e., Gram negative bacteria. The physico-chemical parameters of the formulation were also optimal with no signs of irritation.

Keywords: Formulation, Evaluation, Cinnamomum, Camphora, Bhimseni Kapur.**Corresponding author:****Md. Rageeb Md. Usman,**

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INTRODUCTION:

It is a concentrate of active ingredients, which targets specific skincare concerns, and the ingredients are powerful, and made up of smaller molecules. The level of active ingredients is higher than in a usual face cream, since the heavier oils and ingredients have been done away with. So, while the latter could have around ten per cent of active ingredients, the former has a whopping seventy per cent or more! Serum is a skincare product you can apply to your skin after cleansing but before moisturizing with the intent of delivering powerful ingredients directly into the skin [1-3]. Serum is particularly suited to this task because it is made up of smaller molecules that can penetrate deeply into the skin and deliver a very high concentration of active ingredients. This makes them a great tool for targeting specific skincare concerns, like wrinkles. Goodbye, signs of aging [4-5].

The present project is also a humble attempt in the same direction, where *Cinnamomum Camphora* has been evaluated microbiologically to check its anti-Procurement of Raw Material

microbial property against various infectious problems. *Cinnamomum Camphora* is commonly used in Indian traditional medicine as antibacterial, antifungal, antioxidant, wound healing, anti-inflammation.

It is reported that the *Cinnamomum Camphora*. has antibacterial activity against some Gram-negative bacteria. Therefore, this study aims to evaluate antibacterial properties of *Cinnamomum Camphora* on the bacteria associated with the skin [6-7].

MATERIAL AND METHOD:

Cinnamomum Camphora Wood is collected in Horticultural Garden and (Bhimsen Kapur) sample was collected from the Market. *camphora* leaves were later cut into slices to reveal the tighter inner stem until when ready for use. The analysis of Essential oils is generally derived from one or more plant parts, such as flowers, woods. Fresh *Cinnamomum Camphora* (Bhimsen Kapur) sample was collected from the Market and authenticated by Botanical survey of India Pune.

Table 1: Procurement of Raw Material

| Sr. No. | Ingredients | Suppliers | Role |
|---------|------------------|-------------------------------------|-----------------|
| 1. | Glycerine | Dhruvicameru chem. Pvt. Ltd. | Humectant |
| 2. | Propylene glycol | Dhruvicameru chem. Pvt. Ltd. | Humectant |
| 3. | Xanthum gum | Acuroorganicspvt. Ltd. | Gelling agent |
| 4. | DMDM Hydantoin | Unicarelid,Goyalchem | preservative |
| 5. | Sodium EDTA | Inter MESH Ltd | Chelating agent |
| 6. | Allantoin | Hi tech life sciences,kip chemicals | Healing agent |

Formulation of Active Herbs**Table 2: Formulation of Active Herbs**

| Ingredients | Parts Used | Category | Qty% |
|---|------------|-----------------------------------|------|
| Cinnamomum Camphora  | Leaves | Antioxidants Anti-inflammatory | 3 |
| Cinnamomum Camphora (Pure Bhimseni Kapur)  | | Demulcent Anti- Microbial | 3 |

| | | | |
|---|------|----------------------------------|---|
| Cinnamomum Camphora | Wood | Anti- bacterial Aromatheraphy | 4 |
|  | | | |

Formulation of Base

Table 3: Formulation of Base

| Sr. No. | Ingredients | Quantity for 100% |
|---------|------------------|-------------------|
| 1. | Glycerine | 5% |
| 2. | Propylene glycol | 5% |
| 3. | Xanthum gum | 0.5% |
| 4. | Sodium EDTA | 0.2% |
| 5. | Allantoin | 0.1% |
| 6. | DMDM Hydentoin | 0.15% |
| 7. | Active | 10% |
| 8. | Water | 79.05% |

EXPERIMENTAL WORK

Extraction of *Cinnamomum camphora*

Extraction of Camphor Oil use in Fractional Distillation and Vaccume Distilation (Fig. 1-3).

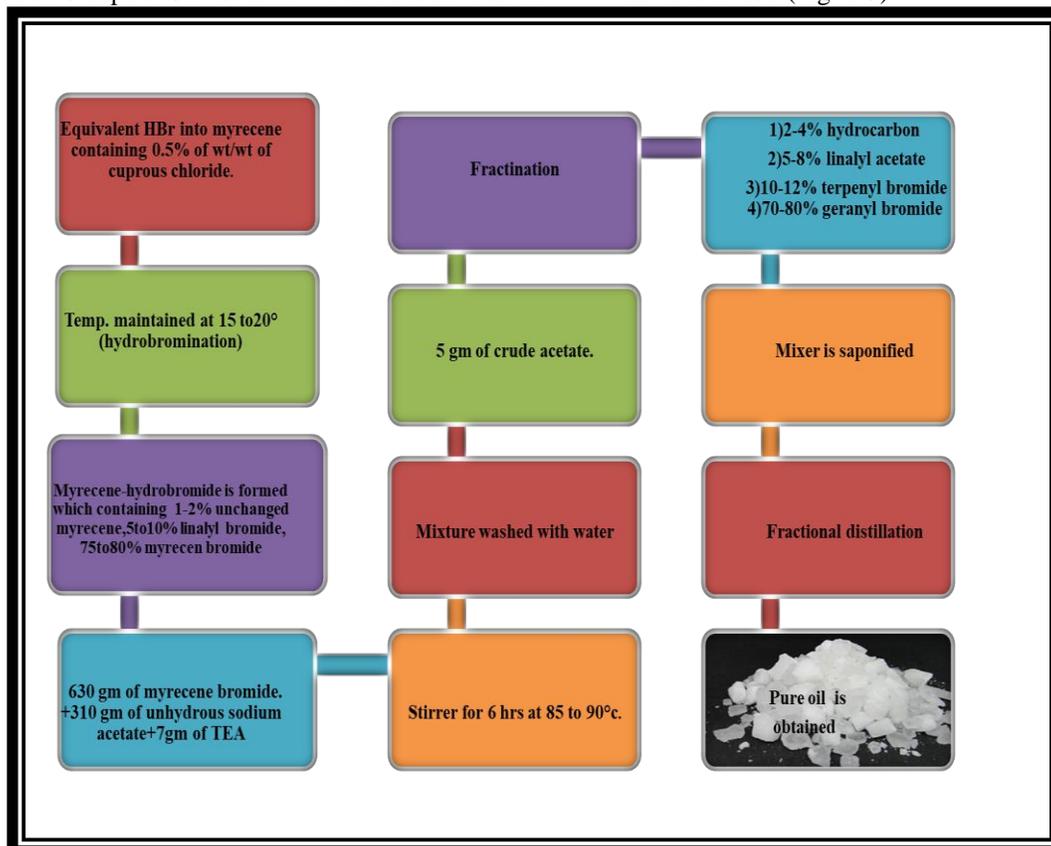


Fig. 2: Extraction of Camphor oil

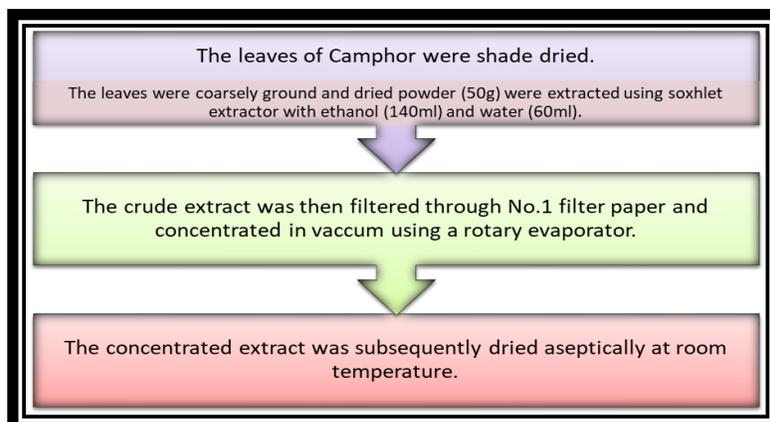


Fig. 3: Extraction of Camphor oil From Leaves

Extraction of Camphor oil Use in Hydrodistillation

Hydrodistillation Method

Hydrodistillation is a traditional method for the extraction of bioactive compounds from plants. In this method, plant materials are packed in a still compartment then water is added in sufficient amount and brought to a boil. Alternatively, direct steam may also be injected into the plant sample [8].

Preparation of Serum

All the ingredients were weighed according to the different percentage listed (Table). The net weight of

all formulated serum was 100g. Add water and EDTA into a disinfected glass beaker and stir, until EDTA has dissolved. Add hyaluronic acid and mix thoroughly with a stick blender or homogenizer until phase A is free of lumps. Add phase B to phase A, stir well after each ingredient has been added. Mix with stick blender. Add phase C to phase A/B, again, stirring well after each ingredient. Especially after sodium acrylate uses the stick blender. Serum should be free of any lumps. Viscosity can be adjusted by adding, 2.5% of the sodium acrylate, if needed [9-12].

Phytochemical Screening

Table 4: Preliminary Phytochemical Analysis of Leaves of Camphor

| Phytochemical Constituents | Leaves Extract | | |
|----------------------------|----------------|------------|--------|
| | Ethanol | Chloroform | Hexane |
| Alkaloids | + | + | - |
| Flavonoids | + | + | + |
| Terpenoids | + | + | - |
| Tannins | - | + | - |
| Saponins | + | - | - |
| Carbohydrate | + | + | - |

+ = indicates presence of phytochemicals, - = indicates absence of phytochemicals

Table 5: Preliminary Phytochemical Analysis of Wood of Camphor

| Sr. No. | Chemical Constituents | <i>Cinnamomum camphora</i> (Wood Extract) |
|---------|-----------------------|---|
| 1 | Resin | - |
| 2 | Tannins | + |
| 3 | Anthraquinones | - |
| 4 | Treprenoids | + |
| 5 | Flavonoids | + |
| 6 | Alkaloids | - |
| 7 | Carbohydrates | + |
| 8 | Saponin | - |

+ = indicates presence of phytochemicals, - = indicates absence of phytochemicals

Antioxidant Activity of Extracts**1) Reducing Power Method**

The reducing power was assayed by taking different concentration of extract (1ml) from each other were mixed in different test tubes with 2.5 ml of phosphate buffer (pH-7) and 2.5 ml of 1% potassium ferric

cyanide. The mixture was then incubated at 50°C for 20 minutes. Then 2.5 ml of trichloroacetic acid (10%) solution was added to the mixture, which was mixed for 15 minutes. Finally 1.25 ml of distilled water was mixed with 0.50 ml of FeCl₃ solution (0.1 w/v). The absorbance was measured at 700nm.

Table 6: Determination of Antioxidant Activity of Extract

| Sr. No. | Extract | Concentration (µg/ml) and absorbance | |
|---------|-------------------------|--------------------------------------|------|
| | | 10 | 20 |
| 1 | Cinnamomum Camphora Oil | 0.18 | 0.96 |
| | | | |

Microbial Assay**Procurement of organisms**

The standard bacterial culture used for this study was procured from our college.

Preparation of media

Weighed a medium in required quantity and added to a dry conical flask.

Added required quantity of freshly distilled water, a little at a time with constant agitation to prevent the formation of lumps.

Heated the flask in boiling water bath with intermittent stirring till the contents were completely dissolved. Filled the test tubes using a funnel when medium is hot.

Plugged the test tubes of medium using non-absorbent cotton, covered them with paper and tie them

together using a rubber band sterilized for 15 minutes at 121°C i.e. at 15 ibsp.s.i. pressure [13-15].

Method Used for Evaluation of Microbial Activity of *Cinamomum camphora***1. The Well Diffusion Method**

- The Petri dishes prepared in the above step are used in this method.
- After the media solidified small well (3-4mm) bored on the media with the help of sterilized cork borer.
- Different concentration of extract prepared.
- These well were filled with different concentration of extract of *Cinamomum Camphora* and incubated at 34°C for 24hrs.
- Tetracycline solution used as standard.
- After 24 hrs. zone of inhibition was observed 10.

Table No.7 concentration of extract

| Sr. No. | Concentrations | Extract |
|---------|----------------|---------|
| 1. | 1500mg | 1.5ml |
| 2. | 2000mg | 2ml |

Observation

Following plate No.1 & 2 showing measurement of zone of inhibition when using well diffusion method for *Propionibacterium acnes*.

Table 8: Diameter of Zone of Inhibition of *Cinamomum camphora* Extract

| Microorganism | Diameter of zone of inhibition in(mm) | |
|-------------------------|---------------------------------------|-----------|
| | Ethanol extract | |
| Propionibacterium acnes | 1500mg/ml | 2000mg/ml |
| | 14mm | 15mm |

Evaluation Tests for Product

Physical Appearance

- **Colour-** White to Yellowish
- **Odour-** Aromatic/Pleasant
- **Consistency-** Liquid
- **Homogeneity-** The formulation produces uniform distribution of extract.

1. Thermal Stability

Formulation and development of a pharmaceutical product is not complete without proper stability analysis carried out on it to determine physical and chemical stability and thus safety of the product. A general methodology for predicting the stability is accelerated stability analysis which subjects the material to elevated temperatures. The stability studies were carried out as per ICH guidelines. Short term accelerated stability study was carried out for the period of 3 months for the formulation. The samples were stored at different storage conditions of temperatures such as 3-5°C, 25°C RH=60% and 40°C±2% RH=75%. Samples Were withdrawn on monthly interval and analysed

| Stability | 1 st week | 2 nd week | 3 rd week | 4 th week |
|-----------|----------------------|----------------------|----------------------|----------------------|
| A | Stable | Instable | Instable | Instable |
| B | Stable | Stable | Instable | Instable |
| C | Stable | Stable | Stable | Stable |
| D | Stable | Stable | Stable | Instable |

CONCLUSION

From the above test it was observed that formulation C was stable throughout stability period, formulation A, B, and D shown instability at week 3 week 4.

2. pH Test

The pH meter was calibrated using standard buffer solution. About 1 ml of the serum was weighed and dissolved in 50.0 ml of distilled water and its pH was measured. pH of prepared serum is 5.2.

3. Spreadability

Spread ability denotes the extent of area to which the serum readily spreads on application to skin or the affected part. To simulate human skin, Fisher brand filter Paper was chosen. Each filter paper weighs within milligrams of any other sheet of that size and type. A Becton Dickinson & Co. 5ml latex syringe without the needle attached was used. Liquid pushed out of the needle attachment end of the B-D syringe formed very uniform drops. Each drop is approximately 0.03 gram in weight. Standard aluminium foil is used as a base to lay the filter paper on for testing. The prepared serum has excellent spreading property.

Organoleptic, Physicochemical and Stability Testing of Serum

1. Homogeneity

This will be confirmed by spread some of the serum formulation on the transparent glass and observe it. The formulation should produce uniform distribution of serum.

2. Rheological Study

Viscosity of the formulation is determined by Brookfield® Viscometer at 100rpm, using spindle type model S64.5 ml of the serum. The serum will be placed in a big mouth container with the spindle dipped in it for about 5 minutes before the measurement.

In vivo studies

Patch test

Patch test was performed on sensitive part of skin, e.g. bend of elbow, popliteal space of skin behind ears. The cosmetic was tested by applying to an area of 1 sq.cm of the skin. Central patches were also applied. The site of the patch was inspected after 24 hours. There was no reactions and then test was repeated once more on the same side. Since there was no reaction as the person was considered as not hyper sensitive and product pass the test [16-18].

Table 9 : Patch Test of Serum

| Sr. No. | Parameter | M1 | M2 | M3 |
|---------|--------------------------------------|------|------|------|
| 1 | Immediately after removal of product | N.R. | N.R. | N.R. |
| 2 | After 24 hrs | N.R. | N.R. | N.R. |
| 3 | After 48 hrs | N.R. | N.R. | N.R. |

EVALUATION PARAMETERS

Table 10: Evaluation Parameter of Essential oil

| Sr. No | Parameters | Observation |
|--------|--------------|----------------|
| 1 | Color | Yellowish |
| 2 | Odor | Aromatic |
| 3 | Consistency | Good |
| 4 | pH | pH 5.2 |
| 5 | Viscosity | 293.15-343.15* |
| 6 | Irritability | Non-irritant |

RESULTS AND DISCUSSION:

The experiment was run at different parameters condition in order to identify the most optimize condition to obtain a maximum yield of essential oil.

Optimization of Oil Extraction

In order to optimize the extraction operating conditions for achieving maximum oil yield, the distillation was conducted at three unlike parameters conditions which were at different water to Raw material ratio (v/w)(8:1, 10:1, 12:1), different extraction period (30min, 60min, 90min, 120min, 150min) and different operating power (200W, 250W).

Analysis of Sample

The extracted essential oils was dried over anhydrous sodium sulfate, weighed and stored in amber vials at 4 °C for the use of analysis. The amount of yield obtained from the extraction was analyzed to evaluate the performance of MAHD in Cinnamomum Camphora extraction. As the result the yield of oil that obtained for every run was calculated by using

$$\text{Yield of essential oil (\%)} = \frac{\text{Amount of essential oil (c) obtained}}{\text{Amount of raw materials (c) used}} \times 100\%$$

The extracted essential oil samples were analyzed by using Gas Chromatography Mass Spectrometry (GC-MS) Agilent 6890 gas chromatography instrument coupled to an Agilent 5973 mass spectrometer and an Agilent Chem in order to identify their chemical constituents. This is an essential method to evaluate the quality of the oil samples. The following operating parameters were used for Cinnamomum Camphora oil sample: capillary GC column HP-5MS 5% phenylmethylsiloxane (30 x 0.25 mm i.d. x 0.25 mm film thickness), a carrier gas Helium (flow rate

1.2 mL min⁻¹) and a split-less injection mode. Injector temperature is 250°C, Oven temperature will be set initially at 50°C, and then will be raised to 250°C at a 10°C min⁻¹ rate till the end of analysis. The eluted analytes detected using (5973 network) mass selective detector and Electron Impact ionization (EID) will be carried out at 70 eV. While, Cinnamomum Camphora oil sample, the operating parameters of GC-MS was as followed: system operating in EI mode (70 eV), equipped with a split/splitless injector (280 °C, split ratio 1:20), using DB-5 column (30 x 0.25 mm i.d x 0.25 mm). The temperature program was 50°C (5 min) rising to 300°C. at rate of 5°C/min. Injector and detector temperature was 280°C. Helium was used as carrier gas at a flow rate 1 mL/min

Effect of Extraction Time on Yield

The yield of oil extracted from Cinnamomum Camphora at different extraction time in a fix microwave power of 250W and water to raw material ratio of 8:1. From the graph, the amount of yield does not change significantly after 90 minutes. Most of the oil is extracted within 30 to 90 minutes with the yield of 0.61%, 0.80% and 0.85% respectively. Microwave assisted hydro distillation (MAHD) reach the highest yield of 0.85% w/w when extraction time was 90 minutes. However, further increase in extraction time resulted in no improvement in the extraction performance. Similar observations were also reported for MAHD of *Saturejahortensis* and *Satureja Montana* by. The extraction was fast at the beginning of the extraction but get slow gradually by time because when the raw material is exposed to the heat, the plant cell started to degrade and as a consequence the essential oil is released to the environment. However prolong the extraction time caused over heat supplied to the plant material and this lead to the evaporation of the volatile component in the oil.

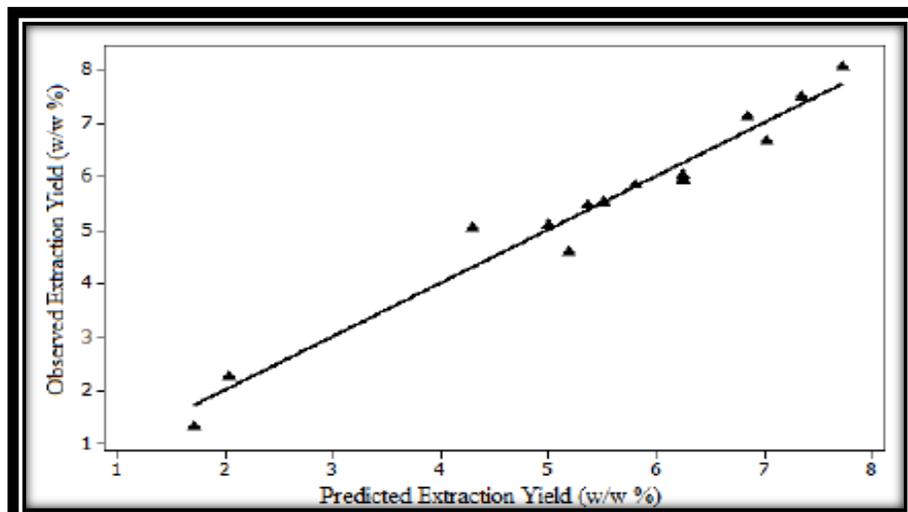


Fig. 6: Extraction Yield of Essential Oil from *Cinnamomum camphora* at Different Extraction Time by 250W in Raw Material to Water Ratio

The yield of oil extracted from *Cinnamomum Camphora* at different extraction time in a fix microwave power of 250W and water to raw material ratio of 8:1. From the graph, the amount of yield does not change significantly after 90 minutes. Most of the oil was extracted within 30 to 90 minutes with the yield of 0.75%, 1.29% and 1.37% respectively.

The solubility of oils in n-hexane was $0.00407 \pm 0.25 \text{ mg/ml}$, $0.00185 \pm 0.45 \text{ mg/ml}$ in methanol, $0.00084 \pm 0.67 \text{ mg/ml}$ in PBS. The Partition coefficient ($K_{o/w}$) for oils was 3.6. From this value, it was shown that given drug comprised of about sufficient lipophilicity that is beneficial to develop the topical drug.

Determination of Geraniol and Ginger Solubility and Partition Coefficient ($K_{o/w}$)

Fractional Distillation Method
Result Obtained by Fractional Distillation is shown in Table below

Table 11: Weight of Oil With Respect to Time

| Weight of oil (g) | Time (mins) |
|-------------------|-------------|
| 0.2 | 240 |
| 0.3 | 480 |
| 0.7 | 720 |
| 0.72 | 960 |
| 0.78 | 1200 |

The amount of pure Camphor oil obtained by extraction method was 2.7g of essential oil per 100g of dry Leaves sample. This gave 2.07% yield of essential oil per 100g of dry *Cinnamomum Camphora*. The temperature used was 78°C i.e. the boiling point of ethanol. The volume of oil was measured at every 4hr interval to determine the oil yield at varying time. As the time increases the Ethanol solvent reduces thereby leaving the oil in the mixture.

Soxhletion Method

Result Obtained by is Shown in Table below

Table 12: Weight of Oil With Respect to Time

| Weight (g) | Time (mins) |
|------------|-------------|
| 0.31 | 240 |
| 0.41 | 480 |
| 0.55 | 720 |
| 0.58 | 960 |
| 0.70 | 1200 |

The oil produced by Vacuum Distillation Method is 2.55g weight of oil per 130g of dry Cinnamomum Camphora sample thereby producing 1.96% oil yield at 780C

6.7 Hydro Distillation Method

Table 13: Weight of Oil with Respect to Time

| Weight (g) | Time (mins) |
|------------|-------------|
| 0.10 | 240 |
| 0.14 | 480 |
| 0.26 | 720 |
| 0.35 | 960 |
| 0.38 | 1200 |

The result of hydro distillation process was 1.23g per 130g of dried Cinnamomum Camphora sample giving 0.95% yield of oil.

Table 14: Result of Essential Oil Extraction

| Method of extraction | % Yield |
|--------------------------------|---------|
| Fractional Distillation Method | 2.07 |
| Soxhletion Method | 1.96 |
| Hydro distillation | 0.95 |

Physical and Chemical Properties of Oil

The essential oil produced was pale yellow, with an aromatic camphoraceous odor, pungent and cooling taste. Because of its high volatility, it was stored in an air-tight container protected from light in cool place. The essential oil is insoluble in water, miscible in alcohol and in oil.

CONCLUSION:

From above discussion it is concluded that Cinnamomum Camphora Extract had antimicrobial property against *Propionibacterium acnes*. From the above experimental work, the Cinnamomum Camphora Extract showing good activity against *Propionibacterium acnes*.

Finally, it was concluded that extract of Cinnamomum Camphora shows antibacterial activity against selected microorganism with increase in concentration the activity is increase therefore it can be incorporated in cosmetics products.

Fractional distillation, Soxhletion and hydrodistillation methods are effective and efficient means of extracting Serum. Extraction is the most common and most economically technique for extracting Natural oil in modern Herbal industry because of its simplicity.

There is high demand for essential oils for various purposes such as medicinal, perfumery, aromatherapy, cosmetic, soap making, insecticides to mention but a few. Imported essential oils are very

expensive to meet the demand of our local consumer industries, therefore it becomes necessary to source and synthesis these oils from local sources, in particular.

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