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**Research Article** 

# DEVELOPMENT AND VALIDATION OF ANALYTICAL METHOD FOR SIMULTANEOUS ESTIMATION OF ECONAZOLE AND TRIAMCINOLONE BY RP- HPLC

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Abstract:		
A new, simple, rapid and precise reverse	phase high performance liquid chromato	graphic method has been developed
for the validation of Econazole and Trie	amcinolone in its pure form as well as	in combined marketed formulation.
Chromatography was carried out on a	Phenomenex Luna C18 (4.6mm×250mm	a) 5µm particle size column using a
mixture of Methanol: Phosphate Buffer		
detection was carried out at 275nm. Th	e retention time of the Econazole and T	Triamcinolone was found to be was
$2.133$ , $3.692 \pm 0.02$ min respectively. The		
accuracy, precision, specificity and robu		
20-60mg/ml of Econazole and 10-30mg/		
be within limits. The method precision fe	÷ •	> 2.0%RSD. The method is useful in
the quality control of bulk and pharmace		
Keywords: Econazole and Triamcinolor	e, RP-HPLC, Validation, Accuracy, Pre	ecision.

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# **INTRODUCTION:**

Analytic method development and validation are key elements of any pharmaceutical development program. HPLC analysis method is developed to identify, quantity or purifying compounds of interest. This technical brief will focus on development and validation activities as applied to drug products.

### Method development:

Effective method development ensures that laboratory resources are optimized, while methods meet the objectives required at each stage of drug development. Method validation, required by regulatory agencies at certain stages of the drug approval process, is defined as the "process of demonstrating that analytical procedures are suitable for their intended use" [1-2]. Understanding of the physical and chemical characteristics of drug allows one to select the most appropriate high performance liquid chromatography method development from the available vast literature. Information concerning the sample, for example, molecular mass, structure and functionality, pKa values and UV spectra, solubility of compound should be compiled. The requirement of removal of insoluble impurities by filtration, centrifugation, dilution or concentration to control the concentration, extraction (liquid or solid phase), derivatization for detection etc. should be checked. For pure compound, the sample solubility should be identified whether it is organic solvent soluble or water soluble, as this helps to select the best mobile phase and column to be used in HPLC method development.

Method development in HPLC can be laborious and time consuming. Chromatographers may spend many hours trying to optimize a separation on a column to accomplish the goals. Even among reversed phase columns, there is astonishing diversity, owing to differences in both base silica and bonded phase characteristics. Many of these show unique selectivity. What is needed is a more informed decision making process for column selection that may be used before the chromatographer enters the laboratory. The method of column selection presented here involves a minimal investment in time initially, with the potential of saving many hours in the laboratory.

Analytic methods are intended to establish the identity, purity, physical characteristics and potency of the drugs that we use. Methods are developed to support drug testing against specifications during manufacturing and quality release operations, as well as during long-term stability studies. Methods that support safety and characterization studies or evaluations of drug performance are also to be evaluated. Once a stability-indicating method is in place, the formulated drug product can then be subjected to heat and light in order to evaluate the potential degradation of the API in the presence of formulation excipients [3, 4].

The three critical components for a HPLC method are: sample preparation (% organic, pН, shaking/sonication, sample size, sample age) analysis conditions (%organic, pH, flow rate, temperature, wavelength, and column age), and standardization (integration, wavelength, standard concentration, and response factor correction). During the preliminary method development stage, all individual components should be investigated before the final method optimization. This gives the scientist a chance to critically evaluate the method performance in each component and streamline the final method optimization [5]. The percentage of time spent on each stage is proposed to ensure the scientist will allocate sufficient time to different steps. In this approach, the three critical components for a HPLC method (sample preparation, HPLC analysis and standardization) will first be investigated individually [6-8].

The degraded drug samples obtained are subjected to preliminary chromatographic separation to study the number and types of degradation products formed under various conditions [9]. Scouting experiments are run and then conditions are chosen for further optimization [10]. Resolving power, specificity, and speed are key chromatographic method attributes to keep in mind during method development [11]. Selectivity can be manipulated by combination of different factors like solvent composition, type of stationary phase, mobile phase, buffers and pH. Changing solvents and stationary phases are the most comfortable approaches to achieve the separation. The proper range of pH is an important tool for separation of ionizable compounds. Acidic compounds are retained at low pH while basic compounds are more retained at higher pH. The neutral compounds remain unaffected. The pH range 4-8 is not generally employed because slight change in pH in this range would result in a dramatic shift in retention time. However, by operating at pH extremes (2-4 or 8-10), not only is there a 10-30 fold difference in retention time that can be exploited in method development but also the method can be made more robust which is a desirable outcome with validation in minutes [12,13]. Various steps for HPLC method development are given below.

### Requirements for good method development: Choosing the appropriate HPLC column:

 $C_{18}$  columns are the commonly used columns in HPLC method analysis.  $C_8$  or Octyl bonded phases are also used occasionally. Like  $C_{18}$ , they are nonpolar, but not as hydrophobic. Therefore, retention times for hydrophobic compounds are typically shorter. Also, they may show somewhat different selectivity than  $C_{18}$  due to increased base silica exposure unique selectivity results in proton interaction of the bonded phase with electron deficient functional groups of solute molecules.

### **Column Dimensions:**

This refers to the length (Figure-3.3) and internal diameter of the packing media bed within the column tube. Short columns (30-50mm) offer short run times, fast equilibration, low back pressure and high sensitivity. Long columns (250-300mm) provide higher resolving power, but create more backpressure, lengthen analysis times and use more solvent. Narrow column (2.1mm and smaller) beds inhibit sample diffusion and produce narrower, taller peaks and a lower limit of detection. They may require instrument modification to minimize distortion of the chromatography. Wider columns (10-22mm) offer the ability to load more sample.

# **Particle shape:**

Most modern chromatographic packings have spherical particles (Figure-3.4), but some are irregular in shape. Spherical particles offer reduced back pressures and longer column life when using viscous mobile phases like 50:50 MeOH: H2O.

# Particle Size:

This refers to the average diameter (Figure-3.5) of the packing media particles. Standard particle sizes range from  $3\mu$ m (high efficiency) to 15-20 $\mu$ m (preparative). A  $5\mu$ m particle size offers a good compromise between efficiency and back pressure. Smaller particles pack into columns with a higher density, allowing less diffusion of sample bands between particles and causing narrower, sharper peaks. However, smaller particles also cause higher solvent back pressures. As a rule of thumb, 1.5 or  $3\mu$ m particle sizes are to chosen for resolving complex, multi-component samples. Otherwise, 5 or 10 $\mu$ m packings should be considered.

### Surface area

Expressed in m2/gram, the total surface area of a particle is the sum of the outer particle surface and the interior pore surface (Figure-3.6). Solute retention is greater on packings that have a high surface area. High surface areas generally provide longer retention, greater capacity and higher

resolution. As a rule of thumb, a base material with maximum surface area is to be used for resolving complex and multi-component samples.

### Pore size:

This refers to the average size of the pores or cavities present in porous packing particles (Figure-3.7). Pore sizes range from 60Å on the low end to greater than 10,000Å on the high end. Larger pores allow larger solute molecules to be retained longer through maximum exposure to the surface area of the particles. A pore size of 150Å or less is chosen for sample MW  $\Box$  2000. For sample with molecular weight greater than 2000, columns with a pore size of 300Å or greater are to be used.

# **Bonding Type:**

This refers to how the bonded phase is attached to the base material (Figure-3.8). Monomeric bonding uses single-point attachment of each bonded phase molecule to the base material. Polymeric bonding uses multi-point attachment of each bonded phase molecule to the base material. Polymeric bonding offers increased column stability, particularly when highly aqueous mobile phases are used. Polymeric bonding also enables the column to accept higher sample loading.

### **Carbon Load:**

Carbon load (Figure-3.9) refers to the amount of bonded phase attached to the base material. For C18, C8 and phenyl packings, the carbon load is a good indicator of hydrophobic retention. Higher carbon loads generally give higher column capacities, greater resolution and longer run times. Conversely, low carbon loads shorten run times and may show different selectivity because of greater exposure of the base material. Choose high carbon loads for complex samples which require the maximum degree of separation. Suitable carbon loads must be selected to give shorter analysis times for simple sample mixtures and for samples which require high water content for solubility or stability.

# End capping:

End capping applies only to reversed phase chromatography and is the process of bonding short hydrocarbon chains to free silanols remaining after the primary bonded phase has been added to the silica base. End capping reduces peak-tailing of polar solutes that interact excessively with the otherwise exposed, mostly acidic silanols. Non-end capped packings provide a different selectivity than do the end capped packings, especially for polar samples. **Detectors:** 

Various detectors used in HPLC instrument include UV-Visible detector, photodiode array detector, fluorescence detector, conductivity detector, refractive index detector, electrochemical detector, mass spectrometer detector and evaporative light scattering detector. UV-Visible detectors are typical in many laboratories as they can detect a wide array of compounds

# pH range:

Method development within the different pH ranges from 1 to 12 for better chromatographic resolution between two or more peaks of an analyte depends upon three main factors, column efficiency, selectivity and retention time. The ionizable analytes are either bases or acids and they affect the above three factors dramatically with change in pH. Retention time can be improved by changing the pH that will lead to easy separation of ionizable analytes from non-ionized forms. Change in the mobile phase pH can also improve column efficiency because it alters both the ionization of the analyte and the residual silanols and it also minimizes secondary interactions between analytes and the silica surface that lead to poor peak shape. To achieve optimum resolution, it requires change in the pH of mobile phase. Method development can proceed by investigating parameters of chromatographic separations first at low pH and then at higher pH until optimum results are achieved.

### **MATERIALS AND METHODS:**

Econazole /Triamcinolone-Sura labs,Water and Methanol for HPLC-LICHROSOLV (MERCK), Acetonitrile for HPLC-Merck, Potassium Dihydrogen Phosphate- Finar Chemicals.

### Hplc method development: Trails:

### **Preparation of standard solution:**

Accurately weigh and transfer 10 mg of Econozole & Triamcinolone working standard into a 10ml of clean dry volumetric flasks add about 7ml of Methanol and sonicate to dissolve and removal of air completely and make volume up to the mark with the same Methanol. Further pipette 0.1ml of the above Econozole and 0.3ml of the Triamcinolone stock solutions into a 10ml volumetric flask and dilute up to the mark with Methanol.

### **Procedure:**

Inject the samples by changing the chromatographic conditions and record the chromatograms, note the conditions of proper peak elution for performing validation parameters as per ICH guidelines.

### Mobile Phase Optimization:

Initially the mobile phase tried was Methanol: Water and Water: Acetonitrile and Methanol: Phosphate Buffer: ACN with varying proportions. Finally, the mobile phase was optimized to Acetonitrile: Phosphate Buffer in proportion 45:55 v/v respectively.

## **Optimization of Column:**

The method was performed with various columns like C18 column, Symmetry and Zodiac column. Phenomenex Luna C18 ( $4.6 \times 250$ mm, 5µm) particle size was found to be ideal as it gave good peak shape and resolution at 1ml/min flow.

# OPTIMIZED CHROMATOGRAPHIC CONDITIONS:

Instrument used : Waters HPLC with auto											
sampler and PDA Detector 996 model.											
Temperature :	35°C										
Column :	Phenomenex Luna C18										
(4.6×250mm, 5µm) particl	e size										
Buffer	: Dissolve 6.8043										
of potassium dihydrogen p	hosphate in 1000 ml HPLC										
water and adjust the	pH 4.6 with diluted										
orthophosphoric acid. Filte	er and sonicate the solution										
by vacuum filtration and u	ltra sonication.										
рН	: 4.6										
Mobile phase	: Acetonitrile:										
Phosphate Buffer (45:55 v/v)											
Flow rate	: 1ml/min										
Wavelength	: 245 nm										

#### VALIDATION:

Injection volume :

Run time

#### **PREPARATION OF BUFFER AND MOBILE PHASE:**

10 µl

7 min

•

**Preparation of Potassium dihydrogen Phosphate** (**KH2PO4**) **buffer (pH-4.6):** Dissolve 6.8043 of potassium dihydrogen phosphate in 1000 ml HPLC water and adjust the pH 4.6 with diluted orthophosphoric acid. Filter and sonicate the solution by vacuum filtration and ultra-sonication.

**Preparation of mobile phase:** Accurately measured 450 ml (45%) of Methanol, 550 ml of Phosphate buffer (55%) were mixed and degassed in digital ultra sonicator for 15 minutes and then filtered through 0.45  $\mu$  filter under vacuum filtration.

**Diluent Preparation:** The Mobile phase was used as the diluent.

### **RESULTS AND DISCUSSION:**

### **Optimized Chromatogram (Standard)**

Mobile phase ratio	: Methanol:
Phosphate Buffer (pH-4.2)	(37:63 v/v)
Column	: Phenomenex Luna C18
(4.6mm×250mm) 5µm par	ticle size
Column temperature	: 35°C
Wavelength	: 275nm
Flow rate	: 1ml/min
Injection volume	: 10µl
Run time	: 6minutes

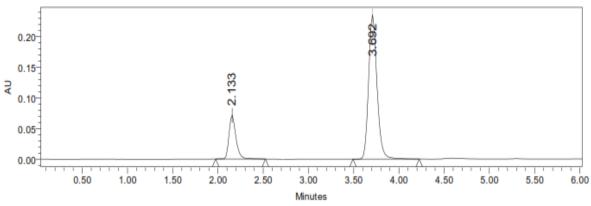


Figure-: Optimized Chromatogram (Standard)

	Table-: Optimized Chromatogram (Standard)												
S.No.	Name	RT	Area	Height	USP Tailing	USP Plate Count	Resolution						
1	Triamcinolone	2.133	526389	86756	1.56	5679							
2	Econazole	3.692	1687285	367532	1.79	8685	9.8						

**Observation**: From the above chromatogram it was observed that the Econazole and Triamcinolone peaks are well separated and they show proper retention time, resolution, peak tail and plate count. So, it's optimized trial.

### **Optimized Chromatogram (Sample)**

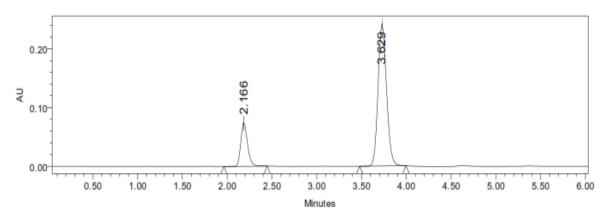


Figure-: Optimized Chromatogram (Sample)

S.No.	Name	Rt	Area	Height	USP Tailing	USP Plate Count	Resolution
1	Triamcinolone	2.166	536587	77464	1.57	5789	
2	Econazole	3.629	1695846	378564	1.80	8795	10.01

**Table-: Optimized Chromatogram (Sample)** 

- Resolution between two drugs must be not less than 2.
- Theoretical plates must be not less than 2000.

# System Suitability:

# Table-: Results of system suitability for Triamcinolone

S.No.	Peak Name	RT	Area (µV*sec)	Height (µV)	USP Plate Count	USP Tailing
1	Triamcinolone	2.152	526358	86598	5695	1.56
2	Triamcinolone	2.157	526548	86254	5652	1.57
3	Triamcinolone	2.141	526854	86598	5627	1.56
4	Triamcinolone	2.133	526598	86245	5692	1.57
5	Triamcinolone	2.166	524874	86521	5641	1.56
Mean			526246.4			
Std. Dev.			787.353			
% RSD			0.149617			

### **Acceptance Criteria:**

- %RSD of five different sample solutions should not more than 2.
- The %RSD obtained is within the limit, hence the method is suitable.

Table-: Results of System suitability for Econazoie											
S.No.	Peak Name	RT	Area (µV*sec)	Height (µV)	USP Plate Count	USP Tailing	Resolution				
1	Econazole	3.674	1682821	1686958	8659	1.56	9.8				
2	Econazole	3.631	1682726	1685745	8675	1.57	9.9				
3	Econazole	3.625	1687361	1685421	8692	1.56	9.8				
4	Econazole	3.692	1682811	1685242	8642	1.57	9.8				
5	Econazole	3.629	1683816	1685364	8635	1.58	9.8				
Mean			1683907								
Std. Dev.			1982.03								
% RSD			0.117704								

# Table-: Results of system suitability for Econazole

### Acceptance Criteria:

- %RSD of five different sample solutions should not more than 2.
- The %RSD obtained is within the limit, hence the method is suitable.

Assay (Standard):

S.No	Name	RT	Area	Height	USP Tailing	<b>USP Plate Count</b>	Injection
1	Triamcinolone	2.152	526358	86598	1.56	5698	1
2	Triamcinolone	2.198	526584	86784	1.57	5687	2
3	Triamcinolone	2.179	529658	86253	1.56	5639	3

Table-: Peak results for assay standard of Triamcinolone

# Table-: Peak results for assay standard of Econazole

S.No.	Name	RT	Area	Height	USP Tailing	USP Plate Count	Injection
1	Econazole	3.646	1687589	365879	1.80	8659	1
2	Econazole	3.604	1685987	365854	1.79	8697	2
3	Econazole	3.610	1685974	369854	1.80	8675	3

# Table-: Peak results for Assay sample of Triamcinolone

S.No	Name	RT	Area	Height	USP Tailing	USP Plate Count	Injection
1	Triamcinolone	2.152	536859	87584	1.58	5789	1
2	Triamcinolone	2.150	532654	87965	1.59	5784	2
3	Triamcinolone	2.187	532685	87465	1.58	5769	3

### Table-: Peak results for Assay sample of Econazole

S.No	Name	RT	Area	Height	USP Tailing	USP Plate Count	Injection				
1	Econazole	3.646	1698568	378562	1.81	8759	1				
2	Econazole	3.651	1698574	375847	1.80	8795	2				
3	Econazole	3.601	1698547	376584	1.81	8745	3				
6ASSA	ASSAY =										

Sample area	Weight of standard	Dilution of sample	Purity	Weight of tablet	
X	>	<x< td=""><td>X</td><td>×1</td><td>00</td></x<>	X	×1	00
Standard area	Dilution of standard	Weight of sample	100	Label claim	

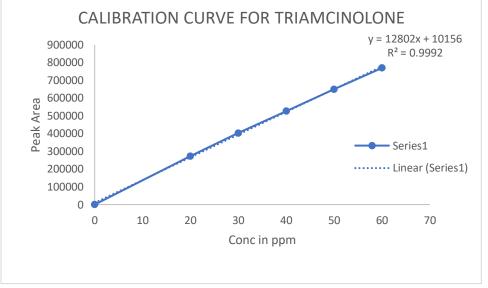
= 99.89%

The % purity of Econazole & Triamcinolone in pharmaceutical dosage form was found to be 99.89%

# LINEARITY:

Chromatographic data for linearity study of triamcinolone: Table-: Chromatographic Data for Linearity Study of Triamcinolone

Concentration	Average
µg/ml	Peak Area
20	272897
30	402986
40	526389
50	649785
60	769287



**Fig-:** Calibration Curve of Triamcinolone

Chromatographic data for linearity study of econazole:
Table-: Chromatographic Data for Linearity Study of Econazole

Concentration µg/ml	Average Peak Area
10	1000237
15	1448768
20	1887285
25	2365897
30	2826845

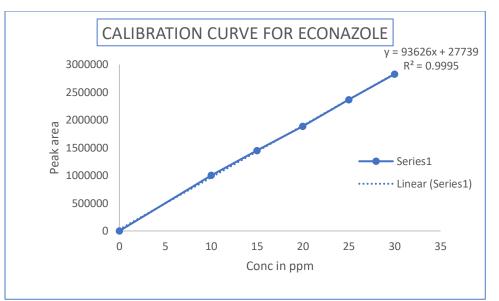


Fig-: Calibration Curve of Econazole

	Table-: Results of repeatability for Triamcinolone:									
S. No.	Deels Marrie	Retention	Area	Height	<b>USP Plate</b>	<b>USP</b> Tailing				
<b>5.</b> INO.	Peak Name	time	(µV*sec)	( <b>µV</b> )	Count					
1	Triamcinolone	2.157	526358	86598	5689	1.56				
2	Triamcinolone	2.159	524856	86542	5687	1.57				
3	Triamcinolone	2.186	526985	86578	5684	1.56				
4	Triamcinolone	2.160	528654	86354	5689	1.56				
5	Triamcinolone	2.170	528457	86958	5639	1.56				
Mean			527062							
Std.dev			1569.114							
%RSD			0.297709							

# **REPEATABILITY:**

# **Acceptance Criteria:**

- %RSD for sample should be NMT 2 •
- The %RSD for the standard solution is below 1, which is within the limits hence method is precise. •

Table-: Results o	of Repeatabili	ty for Econazole:
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S. No.	Peak Name	Retention time	Area (µV*sec)	Height (µV)	USP Plate Count	USP Tailing
1	Econazole	3.603	1687589	367859	8659	1.79
2	Econazole	3.608	1685987	368547	8679	1.80
3	Econazole	3.600	1685987	367985	8645	1.80
4	Econazole	3.696	1685754	365874	8695	1.79
5	Econazole	3.629	1685985	364589	8625	1.79
Mean			1686260			
Std.Dev			749.493			
%RSD			0.044447			

Intermediate precision:

S.No	Peak Name	RT	Area (µV*sec)	Height (µV)	USP Plate count	USP Tailing	%Assay
1	Triamcinolone	2.198	546585	87589	5898	1.58	100%
2	Triamcinolone	2.196	548758	87985	5879	1.59	100%
3	Triamcinolone	2.160	549854	87452	5868	1.58	100%
4	Triamcinolone	2.160	548798	87421	5847	1.59	100%
5	Triamcinolone	2.160	542659	87963	5896	1.58	100%
6	Triamcinolone	2.186	548754	87254	5874	1.59	100%
Mea n			547568				
Std. Dev.			2631.576				
% RSD			0.480593				

Table-: Results of Intermediate precision for Triamcinolone

# Acceptance criteria:

• %RSD of five different sample solutions should not more than 2

	Tuble: Results of Internetiate precision for Leonazore										
S.No.	Peak Name	Rt	Area (µV*sec)	Height (µV)	USP Plate count	USP Tailing	Resolution	%Assay			
1	Econazole	3.623	1698587	385482	8789	1.81	9.8	98%			
2	Econazole	3.611	1698574	385698	8759	1.80	9.8	98.2%			
3	Econazole	3.696	1698532	385748	8754	1.81	9.9	98.7%			
4	Econazole	3.696	1698574	386958	8754	1.81	10.01	99.7%			
5	Econazole	3.696	1698532	385755	5798	1.80	9.98	98.5%			
6	Econazole	3.642	1698547	386558	8762	1.80	10.02	98.2%			
Mean			1698558								
Std. Dev.			23.77113								
% RSD			0.001399								

Acceptance criteria:

• %RSD of five different sample solutions should not more than 2

Table-: Results of Intermediate precision Day 2 for Thankinotone									
S.No.	Peak Name	RT	Area (µV*sec)	Height (µV)	USP Plate count	USP Tailing			
1	Triamcinolone	2.198	536854	8758	5789	1.58			
2	Triamcinolone	2.196	536985	8795	5726	1.59			
3	Triamcinolone	2.178	536587	8746	5742	1.58			
4	Triamcinolone	2.142	532546	8754	5746	1.59			
5	Triamcinolone	2.177	534587	8725	5798	1.58			
6	Triamcinolone	2.177	538598	8726	5785	1.59			
Mean			536026.2						
Std. Dev.			2131.492						
% RSD			0.397647						

**Table-: Results of Intermediate precision Day 2 for Triamcinolone** 

• %RSD of five different sample solutions should not more than 2.

 Table-: Results of Intermediate precision Day 2 for Econazole

S.No.	Peak Name	RT	Area (µV*sec)	Height (µV)	USP Plate count	USP Tailing	Resolution
1	Econazole	3.611	1678598	356875	8875	1.82	9.9
2	Econazole	3.623	1678985	358985	8856	1.83	10.01
3	Econazole	3.684	1678984	358754	8862	1.82	9.9
4	Econazole	3.697	1678985	352412	8849	1.83	10.01
5	Econazole	3.684	1678549	358987	8873	1.82	9.9
6	Econazole	3.684	1678984	358986	8842	1.83	10.01
Mean			1678848				
Std. Dev.			212.8048				
% RSD			0.012676				

• %RSD of five different sample solutions should not more than 2.

# ACCURACY:

Accuracy at different concentrations (50%, 100%, and 150%) was prepared and the % recovery was calculated.

%Concentration (at specification Level)	Area	Amount Added (ppm)	Amount Found (ppm)	% Recovery	Mean Recovery
50%	267011.3	20	20.063	100.315%	
100%	523752.3	40	40.118	100.295%	100.28%
150%	778457.3	60	60.133	100.221%	

# Table-: The accuracy results for Triamcinolone

# Acceptance Criteria:

- The percentage recovery was found to be within the limit (98-102%).
  - Table-: The accuracy results for Econazole

%Concentration (at specification Level)	Area	Amount Added (ppm)	Amount Found (ppm)	% Recovery	Mean Recovery
50%	972876.3	10	10.094	100.94%	
100%	1900122	20	19.998	99.99%	100.48%
150%	2851152	30	30.156	100.52%	

The results obtained for recovery at 50%, 100%, 150% are within the limits. Hence method is accurate. **Robustness** 

### **Table: Results for Robustness triamcinolone**

Parameter used for sample analysis	Peak Area	Retention Time	Theoretical plates	Tailing factor
Actual Flow rate of 1.0 mL/min	526389	2.133	5679	1.56
Less Flow rate of 0.9 mL/min	542685	2.210	5264	1.54
More Flow rate of 1.1 mL/min	526483	2.184	5426	1.52
Less organic phase	516854	2.200	5163	1.57
More Organic phase	506898	2.172	5098	1.51

Acceptance criteria: The tailing factor should be less than 2.0 and the number of theoretical plates (N) should be more than 2000.

### **ECONAZOLE:**

Parameter used for sample analysis	Peak Area	Retention Time	Theoretical plates	Tailing factor
Actual Flow rate of 1.0 mL/min	1687285	3.692	8685	1.79
Less Flow rate of 0.9 mL/min	1725468	4.498	8265	1.68
More Flow rate of 1.1 mL/min	1652847	3.505	8415	1.59
Less organic phase	1687485	4.504	8326	1.62
More organic phase	1674524	3.512	8415	1.63

The tailing factor should be less than 2.0 and the number of theoretical plates (N) should be more than 2000.

### **CONCLUSION:**

- In the present investigation, a simple, sensitive, precise and accurate RP-HPLC method was developed for the quantitative estimation of Econazole and Triamcinolone in bulk drug and pharmaceutical dosage forms.
- This method was simple, since diluted samples are directly used without any preliminary chemical derivatisation or purification steps.
- Econazole was found to be freely soluble in chloroform, soluble in water and in glacial acetic acid, slightly soluble in ethanol and in acetonitrile and practically insoluble in ethyl acetate and in n-hexane. Triamcinolone (hydrochloride) was found to be soluble in organic solvents such as ethanol, DMSO, and dimethyl formamide, soluble in water.
- Methanol: Phosphate Buffer (pH-4.2) (37:63 v/v) was chosen as the mobile phase. The solvent system used in this method was economical.
- The %RSD values were within 2 and the method was found to be precise.
- The results expressed in Tables for RP-HPLC method was promising. The RP-HPLC method is more sensitive, accurate and precise compared to the Spectrophotometric methods.
- This method can be used for the routine determination of Econazole and Triamcinolone in bulk drug and in Pharmaceutical dosage forms.

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### **BIBLIOGRAPHY:**

- 1. Dr. Kealey and P.J Haines, Analytical Chemistry, 1<sup>st</sup>edition, Bios Publisher, (2002), PP 1-7.
- A.BraithWait and F.J.Smith, Chromatographic Methods, 5<sup>th</sup>edition, Kluwer Academic Publisher, (1996), PP 1-2.
- Andrea Weston and Phyllisr. Brown, HPLC Principle and Practice, 1<sup>st</sup> edition, Academic press, (1997), PP 24-37.
- 4. Yuri Kazakevich and Rosario Lobrutto, HPLC for Pharmaceutical Scientists,

1<sup>st</sup>edition, Wiley Interscience A JohnWiley & So ns, Inc., Publication, (2007), PP 15-23.

- Chromatography, (online). URL:http://en.wikipedia.org/wiki/Chromatograp hy.
- Meyer V.R. Practical High-Performance Liquid Chromatography, 4<sup>th</sup> Ed. England, John Wiley & Sons Ltd, (2004), PP 7-8.
- Sahajwalla CG a new drug development, vol 141, Marcel Dekker Inc., New York, (2004), PP 421– 426.
- D. H. Shewiy, E. Kaale, P. G. Risha, B. Dejaegher, J. S. Verbeke, Y. V. Heyden, Journal Pharmaceut. Biomed. Anal, 66, 2012, 11-23.
- M. D. Rockville, General Tests, Chapter 621 Chromatography System Suitability, United States Pharmacopeial Convention (USP), USP 31, 2009.
- FDA Guidance for Industry-Analytical Procedures and Method Validation, Chemistry, Manufacturing, and Controls Documentation, Centre for Drug Evaluation and Research (CDER) and Centre for Biologics Evaluation and Research (CBER), 2000.
- 11. Korany MA, Mahgoub H, Ossama TF, Hadir MM. Application of artificial neural networks for response surface modelling in HPLC method development. J Adv Res, 3, 2012, 53-63.
- 12. Swartz ME, Jone MD, Fowler P, Andrew MA. Automated HPLC method development and transfer. Lc Gc N. Am, 75, 2002, 49-50.
- 13. Snyder LR, Kirkland JJ, Glajach JL. X. In Practical HPLC Methods Development. John Wiley, New York, 295, 1997, 643-712.
- 14. Swartz M, Murphy MB. New Fronties in Chromatography. Am Lab, 37, 2005, 22-27.
- 15. Dolan JW. Peak tailing and resolution. Lc Gc N. Am, 20, 2002, 430-436.
- Chan CC, Leo YC, Lam H. Analytical method validation and Instrument Performance Validation. Vol-I, Wiley Interscince, 2004.