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Review Article

**MULTIPLE EMULSION DRUG DELIVERY SYSTEMS:
FORMULATION STRATEGIES, CHARACTERIZATION, AND
PHARMACEUTICAL APPLICATIONS****Vaibhav S. Sahare¹, Mohd Bilal Sufi², Mudabbirul Haque³**¹Research Scholar, Department of Pharmaceutics, New Montfort Institute of Pharmacy, Ashti, Wardha-442202, Maharashtra, India²Associate Professor, Department of Pharmaceutics, New Montfort Institute of Pharmacy, Ashti, Wardha-442202, Maharashtra, India³Principal, New Montfort Institute of Pharmacy, Ashti, Wardha-442202, Maharashtra, India**Abstract:**

Multiple emulsion drug delivery systems are advanced colloidal carriers designed to encapsulate both hydrophilic and lipophilic drugs within a single formulation, enabling controlled and sustained drug release. The two principal types, water-in-oil-in-water (W/O/W) and oil-in-water-in-oil (O/W/O), improve drug stability, bioavailability, and therapeutic efficacy while reducing dosing frequency and enhancing patient compliance. Formulation performance is influenced by the selection of oils, surfactants, polymers, co-surfactants, and preparation techniques such as two-step emulsification, membrane emulsification, ultrasonication, and high-pressure homogenization. Characterization parameters including droplet size, pH, viscosity, drug content, entrapment efficiency, in vitro drug release, and stability are essential for ensuring formulation quality and performance. Multiple emulsions have demonstrated significant potential in topical, oral, transdermal, ocular, and targeted drug delivery, particularly for antifungal, anti-inflammatory, anticancer, and cosmetic applications. Despite challenges related to physical stability and large-scale manufacturing, continued technological advancements are expected to expand their pharmaceutical and commercial applications.

Keywords: Multiple emulsion, W/O/W emulsion, O/W/O emulsion, Controlled drug release, Drug delivery system, Entrapment efficiency, Pharmaceutical applications, Formulation strategies, Characterization, Sustained release.

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1. INTRODUCTION:

Multiple emulsion systems are advanced colloidal drug delivery systems composed of emulsions within emulsions. Unlike conventional oil-in-water (O/W) or water-in-oil (W/O) emulsions, multiple emulsions possess a multicompartiment structure that allows simultaneous encapsulation of hydrophilic and lipophilic drugs. The two most common types are water-in-oil-in-water (W/O/W) and oil-in-water-in-oil (O/W/O) emulsions. Their unique architecture enables controlled drug release, improved drug stability, and enhanced therapeutic

efficacy, making them attractive carriers for pharmaceutical, cosmetic, and biomedical applications.¹

The increasing demand for advanced drug delivery systems has accelerated research on multiple emulsions because conventional dosage forms often exhibit rapid drug release, poor bioavailability, frequent dosing, and limited patient compliance. Multiple emulsions overcome these limitations by acting as drug reservoirs that regulate drug diffusion through multiple interfaces, thereby providing sustained and targeted drug delivery.²

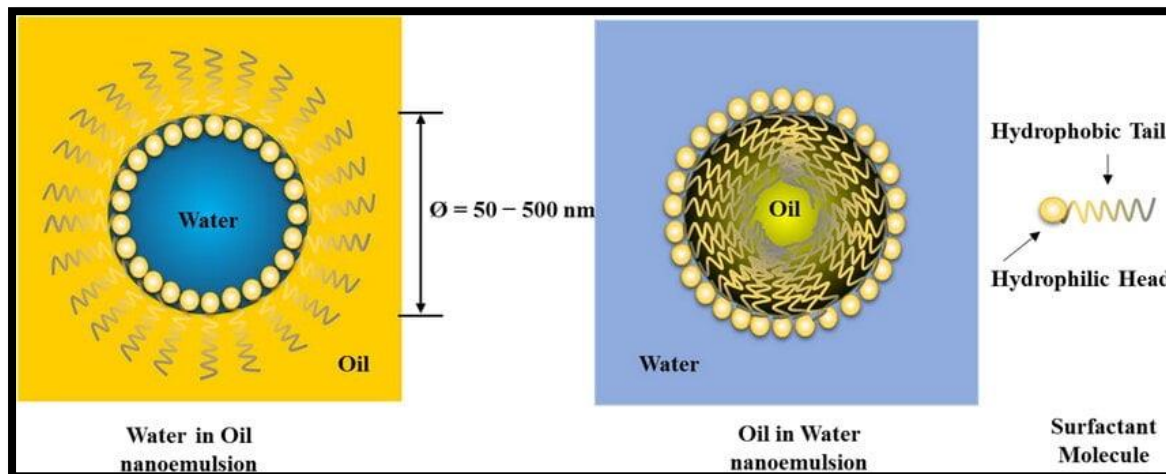


Figure 1. General Structure of Water-in-Oil-in-Water (W/O/W) Multiple Emulsion Drug Delivery System

Multiple emulsions have gained considerable importance in topical drug delivery due to their ability to improve skin hydration, prolong drug residence time, and reduce systemic drug exposure. They have been successfully investigated for the delivery of antifungal, anti-inflammatory, antimicrobial, anticancer, and cosmetic agents. Lipophilic drugs such as econazole nitrate, clotrimazole, ketoconazole, and miconazole are particularly suitable candidates because they exhibit improved retention and sustained release when incorporated into multiple emulsion systems.³

Recent advances in emulsification techniques, surfactant technology, polymer science, and Quality-by-Design (QbD) approaches have significantly improved the stability and reproducibility of multiple emulsions. Modern preparation methods, including high-pressure homogenization, ultrasonication, membrane emulsification, and microfluidics, have facilitated the development of formulations with enhanced encapsulation efficiency and better physicochemical stability. However, challenges such as droplet coalescence, osmotic instability, and large-scale manufacturing continue to limit their commercial application.⁴

This review summarizes the formulation strategies, formulation components, characterization techniques, drug release mechanisms,

pharmaceutical applications, recent advances, and future perspectives of multiple emulsion drug delivery systems. A comprehensive understanding of these aspects may support the development of stable, effective, and commercially viable multiple emulsion formulations for various therapeutic applications.⁵

2. Types of Multiple Emulsion Systems

Multiple emulsions are classified according to the arrangement of the internal and external phases. Among the various configurations, Water-in-Oil-in-Water (W/O/W) and Oil-in-Water-in-Oil (O/W/O) are the most extensively studied because of their versatility in encapsulating drugs with different physicochemical properties. The selection of the emulsion type depends on drug solubility, intended route of administration, desired release profile, and formulation stability.⁶

2.1 Water-in-Oil-in-Water (W/O/W) Multiple Emulsions

The W/O/W multiple emulsion consists of small aqueous droplets enclosed within an oil phase, which is further dispersed in an external aqueous phase. This system is widely used in pharmaceutical formulations because it effectively encapsulates hydrophilic drugs while providing sustained and controlled drug release. The intermediate oil layer acts as a diffusion barrier, reducing rapid drug leakage and improving formulation stability. W/O/W emulsions are particularly suitable for

topical, oral, ocular, and transdermal drug delivery systems.⁷

Advantages of W/O/W Emulsions

- Suitable for hydrophilic drug encapsulation.
- Provides sustained and controlled drug release.
- Protects drugs from chemical degradation.
- Improves topical drug retention.
- Reduces dosing frequency and enhances patient compliance.

2.2 Oil-in-Water-in-Oil (O/W/O) Multiple Emulsions

O/W/O multiple emulsions contain oil droplets dispersed within an aqueous phase, which is subsequently dispersed in a continuous oil phase. These systems are comparatively less explored but are advantageous for the delivery of highly lipophilic drugs and moisture-sensitive compounds. The external oil phase minimizes exposure to aqueous environments, thereby improving the stability of hydrophobic drugs. However, formulation complexity and limited pharmaceutical applications have restricted their widespread use.⁸

Advantages of O/W/O Emulsions

- Suitable for lipophilic drugs.
- Protects moisture-sensitive compounds.
- Reduces hydrolytic degradation.
- Improves oxidative stability.
- Useful for specialized topical and cosmetic formulations.

2.3 Comparison Between W/O/W and O/W/O Multiple Emulsions

The structural arrangement of multiple emulsions determines their pharmaceutical performance. W/O/W systems are generally preferred because they exhibit better patient acceptability, easier

formulation, and broader pharmaceutical applications. In contrast, O/W/O systems are mainly reserved for specialized formulations requiring enhanced protection of lipophilic compounds.⁹

Table 2.1 Comparison Between W/O/W and O/W/O Multiple Emulsions

Parameter	W/O/W	O/W/O
Internal phase	Water	Oil
External phase	Water	Oil
Suitable drugs	Hydrophilic	Lipophilic
Pharmaceutical use	Extensive	Limited
Controlled release	Excellent	Good
Topical application	High	Moderate
Formulation complexity	Moderate	High

2.4 Selection of Appropriate Multiple Emulsion Type

Selection of an appropriate multiple emulsion depends on formulation objectives and the physicochemical characteristics of the active pharmaceutical ingredient. Important considerations include drug solubility, partition coefficient, desired release profile, stability requirements, route of administration, and compatibility of excipients. Proper selection ensures improved encapsulation efficiency, formulation stability, and therapeutic performance.¹⁰

Factors Influencing Selection

- Drug solubility
- Drug partition coefficient
- Desired release profile
- Route of administration
- Surfactant compatibility
- Oil phase characteristics
- Formulation stability

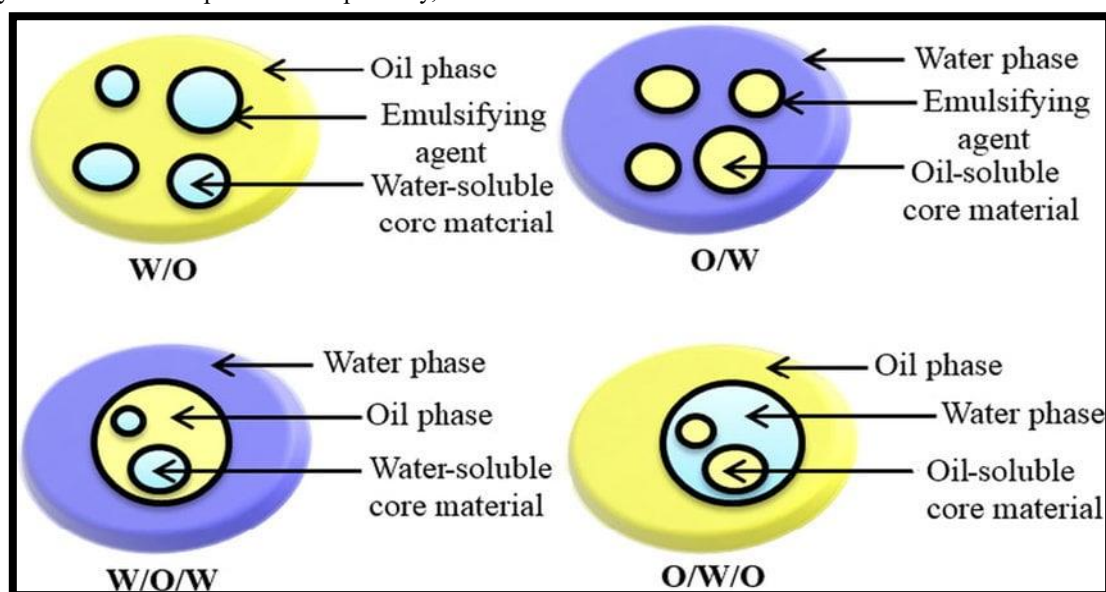


Figure 2. Classification of Multiple Emulsion Systems (W/O/W and O/W/O)

3. Components of Multiple Emulsion Systems

The stability and performance of multiple emulsion systems largely depend on the appropriate selection of formulation components. Each ingredient performs a specific role in maintaining droplet integrity, reducing interfacial tension, enhancing drug encapsulation, and controlling drug release. The major components include the oil phase, aqueous phases, surfactants, co-surfactants, polymers, preservatives, and antioxidants.¹¹

3.1 Oil Phase

The oil phase forms the intermediate barrier between the internal and external aqueous phases in W/O/W multiple emulsions. It controls drug diffusion, influences droplet stability, and contributes to sustained drug release. The oil selected should possess appropriate viscosity, good drug solubilization capacity, chemical stability, and compatibility with surfactants.¹²

Commonly Used Oils

- Liquid paraffin
- Mineral oil
- Medium-chain triglycerides (MCT)
- Olive oil
- Castor oil
- Soybean oil
- Coconut oil
- Isopropyl myristate

3.2 Internal Aqueous Phase

The internal aqueous phase acts as the primary reservoir for hydrophilic drugs. Besides the active pharmaceutical ingredient, this phase may contain buffers, osmotic agents, stabilizers, and viscosity modifiers to improve drug stability and reduce leakage during storage.¹³

Functions

- Drug encapsulation
- Osmotic balance
- Drug stabilization
- Sustained drug release

3.3 External Aqueous Phase

The external aqueous phase serves as the continuous phase of W/O/W emulsions. It provides overall stability and contains hydrophilic surfactants, polymers, preservatives, and viscosity enhancers that minimize droplet aggregation and improve physical stability.¹⁴

3.4 Surfactants

Surfactants are essential for stabilizing both the primary and secondary emulsions. Lipophilic surfactants with low HLB values stabilize the primary W/O emulsion, whereas hydrophilic surfactants with high HLB values stabilize the secondary O/W emulsion. Appropriate surfactant

selection is crucial for preventing coalescence and improving emulsion stability.¹⁵

Lipophilic Surfactants

- Span 20
- Span 60
- Span 80
- Sorbitan trioleate

Hydrophilic Surfactants

- Tween 20
- Tween 40
- Tween 60
- Tween 80

3.5 Co-surfactants

Co-surfactants improve flexibility of the interfacial film, reduce interfacial tension, and facilitate the formation of fine droplets. They also enhance formulation stability and reduce the energy required during emulsification.¹⁶

Examples

- Ethanol
- Propylene glycol
- Polyethylene glycol
- Glycerol
- Transcutol®

3.6 Polymers and Stabilizers

Polymers increase the viscosity of the continuous phase and reduce droplet movement, thereby preventing creaming and coalescence. They also contribute to sustained drug release and improved storage stability.¹⁷

Common Polymers

- Carbopol 934P
- Carbopol 940
- Hydroxypropyl methylcellulose (HPMC)
- Xanthan gum
- Sodium alginate
- Chitosan
- Polyvinylpyrrolidone (PVP)

3.7 Preservatives and Antioxidants

Preservatives prevent microbial contamination, while antioxidants protect oils and active pharmaceutical ingredients from oxidative degradation. Their incorporation is essential for maintaining product quality and extending shelf life.¹⁸

Common Preservatives

- Methyl paraben
- Propyl paraben
- Phenoxyethanol

Common Antioxidants

- Butylated hydroxytoluene (BHT)
- Butylated hydroxyanisole (BHA)
- α -Tocopherol
- Ascorbic acid

Table 3.1 Major Components of Multiple Emulsion Systems

Component	Primary Function	Common Examples
Oil phase	Drug reservoir and diffusion barrier	Liquid paraffin, MCT oil
Internal aqueous phase	Drug loading	Water, phosphate buffer
External aqueous phase	Continuous phase	Purified water
Lipophilic surfactant	Stabilizes W/O emulsion	Span 80
Hydrophilic surfactant	Stabilizes O/W emulsion	Tween 80
Polymer	Viscosity enhancement	Carbopol 934P, HPMC
Preservative	Prevents microbial growth	Methyl paraben
Antioxidant	Prevents oxidation	BHT, α -Tocopherol

4. Formulation Strategies of Multiple Emulsion Systems

The formulation strategy plays a crucial role in determining the stability, encapsulation efficiency, droplet size, and drug release behavior of multiple emulsions. Appropriate selection of emulsification techniques and formulation variables is essential to obtain a stable and reproducible system. Various preparation methods have been developed depending on the physicochemical properties of the drug and the intended pharmaceutical application.¹⁹

4.1 Preparation Methods

The **two-step emulsification method** is the most widely employed technique for preparing multiple emulsions. In this method, a primary emulsion is first prepared and then re-emulsified into a secondary continuous phase to obtain the final

multiple emulsion. Besides this conventional approach, advanced techniques such as phase inversion, membrane emulsification, high-pressure homogenization, and ultrasonication have also been investigated to improve droplet uniformity, encapsulation efficiency, and long-term stability.²⁰

4.2 Factors Affecting Formulation

Several formulation variables influence the quality and performance of multiple emulsions. The type of oil, surfactant concentration, HLB value, phase ratio, homogenization speed, mixing time, viscosity of the continuous phase, and processing temperature collectively determine droplet size, physical stability, and drug release characteristics. Therefore, systematic optimization of these parameters is essential to obtain a stable formulation with prolonged therapeutic performance.²¹

Table 4.1 Common Preparation Methods of Multiple Emulsions

Method	Major Advantage	Limitation
Two-step emulsification	Simple and widely used	Drug leakage may occur
Phase inversion	Uniform droplets	Sensitive to process conditions
Membrane emulsification	Narrow droplet distribution	Costly equipment
High-pressure homogenization	Suitable for industrial production	High energy requirement
Ultrasonication	Rapid emulsification	Possible heat generation

5. Characterization of Multiple Emulsion Systems

Characterization is essential to evaluate the quality, stability, and performance of multiple emulsion formulations. Various physicochemical parameters are assessed to ensure successful drug encapsulation, uniform droplet distribution, and controlled drug release. These evaluations also help in optimizing the formulation and predicting its storage stability and therapeutic efficacy.²²

The commonly evaluated parameters include droplet size, microscopic examination, pH, viscosity, zeta potential, drug content, entrapment efficiency, spreadability, in vitro drug release, and stability studies. Among these, droplet size and entrapment efficiency are considered the most critical quality attributes because they directly influence formulation stability and drug release behavior.²³

Table 5.1 Common Characterization Parameters of Multiple Emulsions

Parameter	Purpose	Instrument/Method
Droplet size	Particle size determination	Optical microscopy/DLS
pH	Skin compatibility	Digital pH meter
Viscosity	Flow behavior	Brookfield viscometer
Drug content	Drug uniformity	UV-Visible spectrophotometer
Entrapment efficiency	Drug encapsulation	Centrifugation method
In vitro drug release	Release profile	Franz diffusion cell
Stability study	Storage stability	ICH stability conditions

6. Pharmaceutical Applications of Multiple Emulsion Systems

Multiple emulsion systems have gained considerable attention in pharmaceutical research because of their ability to provide controlled drug release, improved drug stability, and enhanced bioavailability. Their unique multicompartment structure enables the encapsulation of both hydrophilic and lipophilic drugs, making them suitable for a wide range of therapeutic applications. Recent advancements in formulation technologies have further expanded their use in topical, oral, ocular, transdermal, and targeted drug delivery systems.²⁴

6.1 Topical Drug Delivery

Topical delivery is one of the most widely explored applications of multiple emulsions. These systems improve drug retention on the skin, enhance hydration of the stratum corneum, and provide sustained drug release. Multiple emulsions have been successfully investigated for the treatment of fungal infections, acne, psoriasis, wound healing, and inflammatory skin disorders.²⁵

6.2 Oral and Targeted Drug Delivery

Multiple emulsions have also been investigated for oral controlled-release formulations and targeted drug delivery. Their ability to protect drugs from degradation in the gastrointestinal tract and regulate drug release enhances therapeutic efficacy while reducing dosing frequency. Additionally, they have shown promising potential for delivering proteins, peptides, vaccines, and anticancer agents.²⁶

9. Summary and Conclusion

Multiple emulsion systems have emerged as promising drug delivery platforms because of their unique multicompartment structure, which enables the simultaneous encapsulation of hydrophilic and lipophilic drugs. Compared with conventional emulsions, they offer several advantages, including improved drug stability, enhanced entrapment efficiency, controlled and sustained drug release, and better therapeutic efficacy. Advances in formulation techniques, surfactant selection, and polymer technology have significantly improved the physical stability and reproducibility of these systems. Multiple emulsions have demonstrated broad pharmaceutical applications, particularly in topical, oral, transdermal, ocular, and targeted drug delivery. Despite these advantages, challenges such as thermodynamic instability, droplet coalescence, osmotic imbalance, and scale-up remain important considerations for successful commercialization. Overall, multiple emulsion technology represents a versatile and effective approach for developing novel drug delivery systems, with continued research expected to overcome existing limitations and expand its clinical and industrial applications.

10. Future Scope

Future research on multiple emulsion systems should focus on improving formulation stability,

simplifying manufacturing processes, and facilitating large-scale industrial production. The integration of nanotechnology, Quality-by-Design (QbD), artificial intelligence-based formulation optimization, and advanced emulsification techniques is expected to enhance product quality and reproducibility. The development of biodegradable surfactants, natural polymers, and environmentally sustainable excipients may further improve the safety and biocompatibility of these systems. In addition, multiple emulsions hold significant potential for the delivery of proteins, peptides, vaccines, nucleic acids, herbal bioactives, and personalized medicines. Their application in targeted drug delivery, responsive drug release, and combination therapy is likely to expand with advances in pharmaceutical research. Continued collaboration between academia and industry, supported by regulatory guidance and clinical evaluation, will play a crucial role in translating multiple emulsion technology into commercially successful and patient-friendly therapeutic products.

10. REFERENCES:

1. Florence AT, Whitehill D. The formulation and stability of multiple emulsions. *Int J Pharm.* 1982;11(4):277–308.
2. Schmidts T, Schlupp P, Runkel F. Multiple W/O/W emulsions as novel carrier systems for pharmaceutical and cosmetic applications. *Eur J Pharm Biopharm.* 2009;71(3):339–346.
3. Garti N. Double emulsions—scope, limitations and new achievements. *Colloids Surf A Physicochem Eng Asp.* 1997;123-124:233–246.
4. Muscholik G. Multiple emulsions for food use. *Curr Opin Colloid Interface Sci.* 2007;12(4-5):213–220.
5. Tadros T. *Emulsion Formation and Stability*. Weinheim: Wiley-VCH; 2013.
6. McClements DJ. *Food Emulsions: Principles, Practices, and Techniques*. 3rd ed. Boca Raton: CRC Press; 2015.
7. Jiao J, Rhodes DG, Burgess DJ. Multiple emulsion stability: Influence of processing and formulation variables. *AAPS PharmSciTech.* 2002;3(3):1–8.
8. Benichou A, Aserin A, Garti N. Double emulsions stabilized with macromolecular surfactants. *Adv Colloid Interface Sci.* 2004;108-109:29–41.
9. Ficheux MF, Bonakdar L, Leal-Calderon F, Bibette J. Some stability aspects of multiple emulsions. *Colloids Surf A Physicochem Eng Asp.* 1998;142(2-3):307–312.
10. Schramm LL. *Emulsions, Foams, and Suspensions: Fundamentals and Applications*. Weinheim: Wiley-VCH; 2005.
11. Khar RK, Vyas SP, Ahmad FJ, Jain GK. *Lachman/Lieberman's The Theory and Practice*

- of *Industrial Pharmacy*. 5th ed. New Delhi: CBS Publishers; 2021.
12. Aulton ME, Taylor KMG. *Aulton's Pharmaceutics: The Design and Manufacture of Medicines*. 6th ed. Elsevier; 2022.
 13. Sinko PJ. *Martin's Physical Pharmacy and Pharmaceutical Sciences*. 7th ed. Philadelphia: Wolters Kluwer; 2017.
 14. Allen LV, Popovich NG, Ansel HC. *Ansel's Pharmaceutical Dosage Forms and Drug Delivery Systems*. 11th ed. Wolters Kluwer; 2020.
 15. Shah P, Bhalodia D, Shelat P. Nanoemulsion: A pharmaceutical review. *Syst Rev Pharm*. 2010;1(1):24–32.
 16. Gupta A, Eral HB, Hatton TA, Doyle PS. Nanoemulsions: Formation, properties and applications. *Soft Matter*. 2016;12(11):2826–2841.
 17. Solans C, Solé I. Nano-emulsions: Formation by low-energy methods. *Curr Opin Colloid Interface Sci*. 2012;17(5):246–254.
 18. Anton N, Vandamme TF. Nano-emulsions and micro-emulsions: Clarifications of the critical differences. *Pharm Res*. 2011;28(5):978–985.
 19. Tadros TF, Izquierdo P, Esquena J, Solans C. Formation and stability of nano-emulsions. *Adv Colloid Interface Sci*. 2004;108-109:303–318.
 20. McClements DJ, Rao J. Food-grade nanoemulsions: Formulation, fabrication, properties and applications. *Crit Rev Food Sci Nutr*. 2011;51(4):285–330.
 21. Kumar R, Philip A. Modified transdermal technologies: Breaking the barriers of drug permeation via the skin. *Trop J Pharm Res*. 2007;6(1):633–644.
 22. Prajapati ST, Patel CG, Patel CN. Formulation and characterization of topical drug delivery systems: A review. *Int J Pharm Investig*. 2013;3(4):166–174.
 23. Patel RP, Patel G, Baria AH. Formulation and evaluation aspects of topical emulsions and gels. *Int J Pharm Sci Rev Res*. 2011;6(2):83–91.
 24. Schmidts T, Dobler D, Nissing C, Runkel F. Influence of hydrophilic surfactants on the stability of W/O/W multiple emulsions. *J Colloid Interface Sci*. 2010;338(1):184–192.
 25. Florence AT. Pharmaceutical emulsions and multiple emulsions: Current perspectives. *Pharm Sci Technol Today*. 1998;1(2):58–64.
 26. Gupta PK, Pandit JK, Kumar A, Swaroop P, Gupta S. Pharmaceutical applications of multiple emulsions: A review. *Int J Pharm Sci Nanotechnol*. 2010;3(2):1117–1129.